

# Arctic Aerosols: Sources and Composition

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# Aerosol terminology

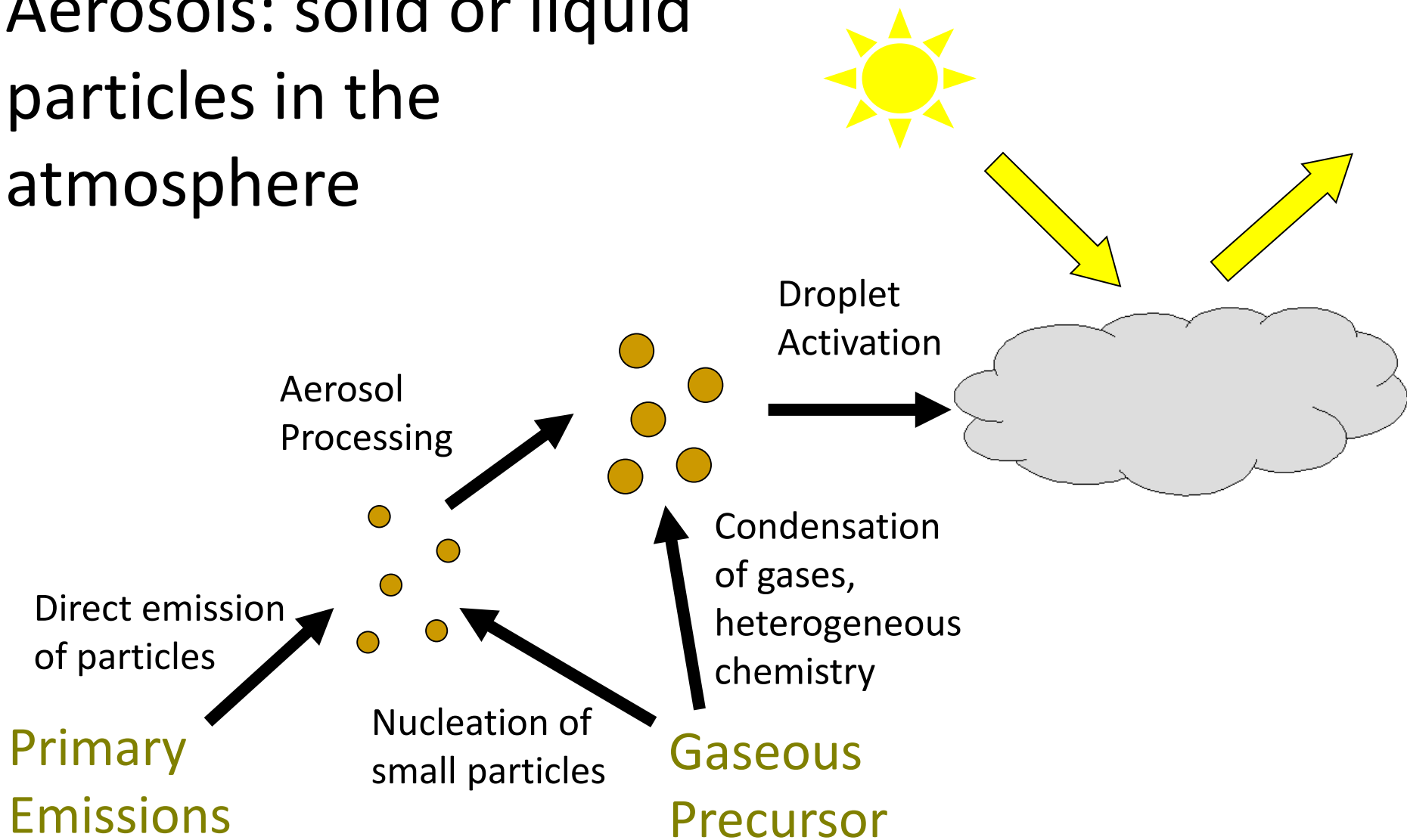
- **Aerosol** – a dispersion of solid and liquid particles suspended in gas (air)

*Note: “aerosol” is defined as the dispersion of both particles and gas, but in common practice it is used to refer to the particles only!*

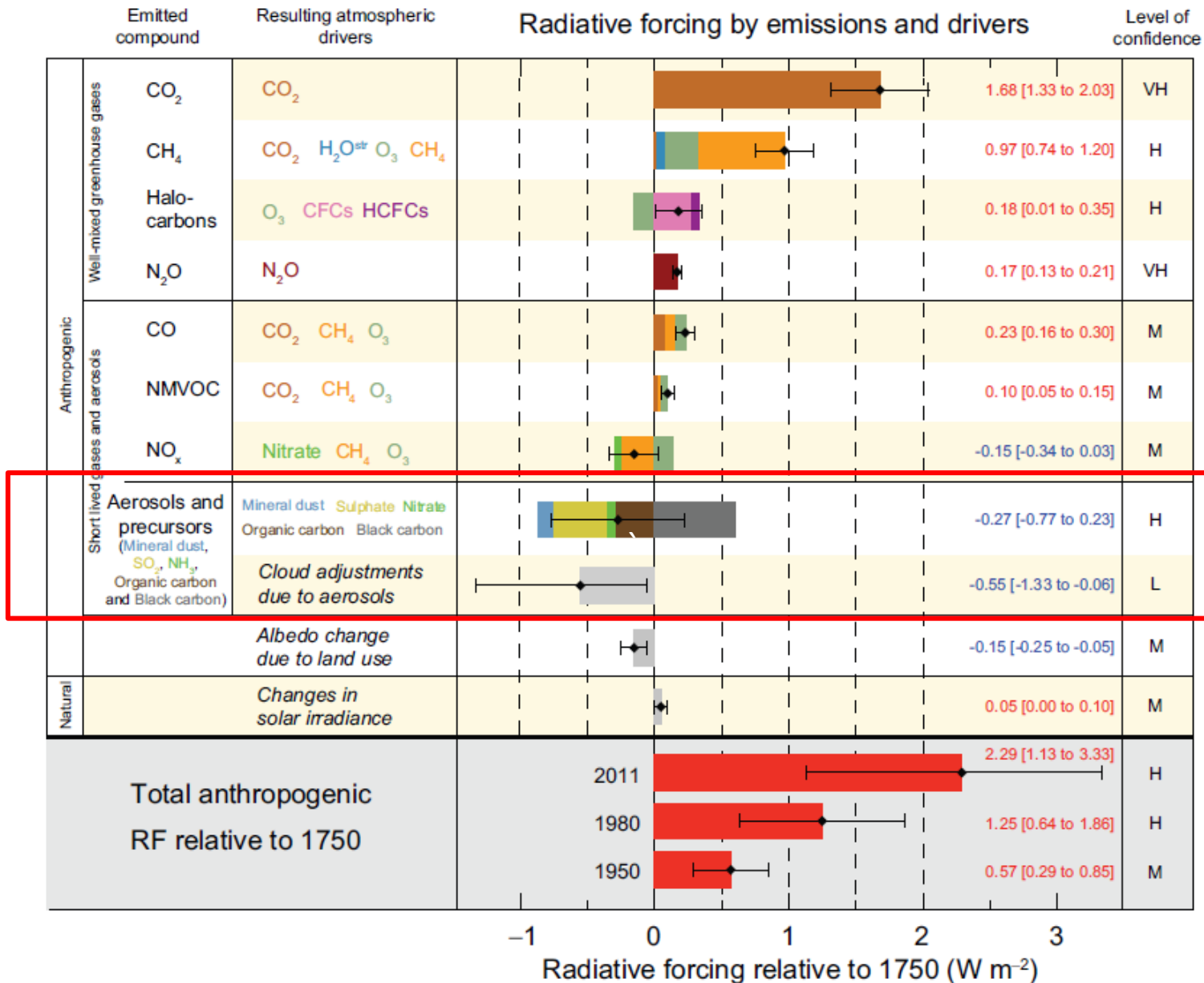
- **Primary aerosol** – atmospheric particles that are emitted or injected directly into the atmosphere.
- **Secondary aerosol** – atmospheric particles that are created by low-volatility gas phase molecules
  - **Condensation** onto existing particles
  - **Nucleation** of new particles

*Primary and secondary may be natural or anthropogenic*

# Aerosols: solid or liquid particles in the atmosphere



# Uncertainties Due to Aerosols



# How much aerosol is there?

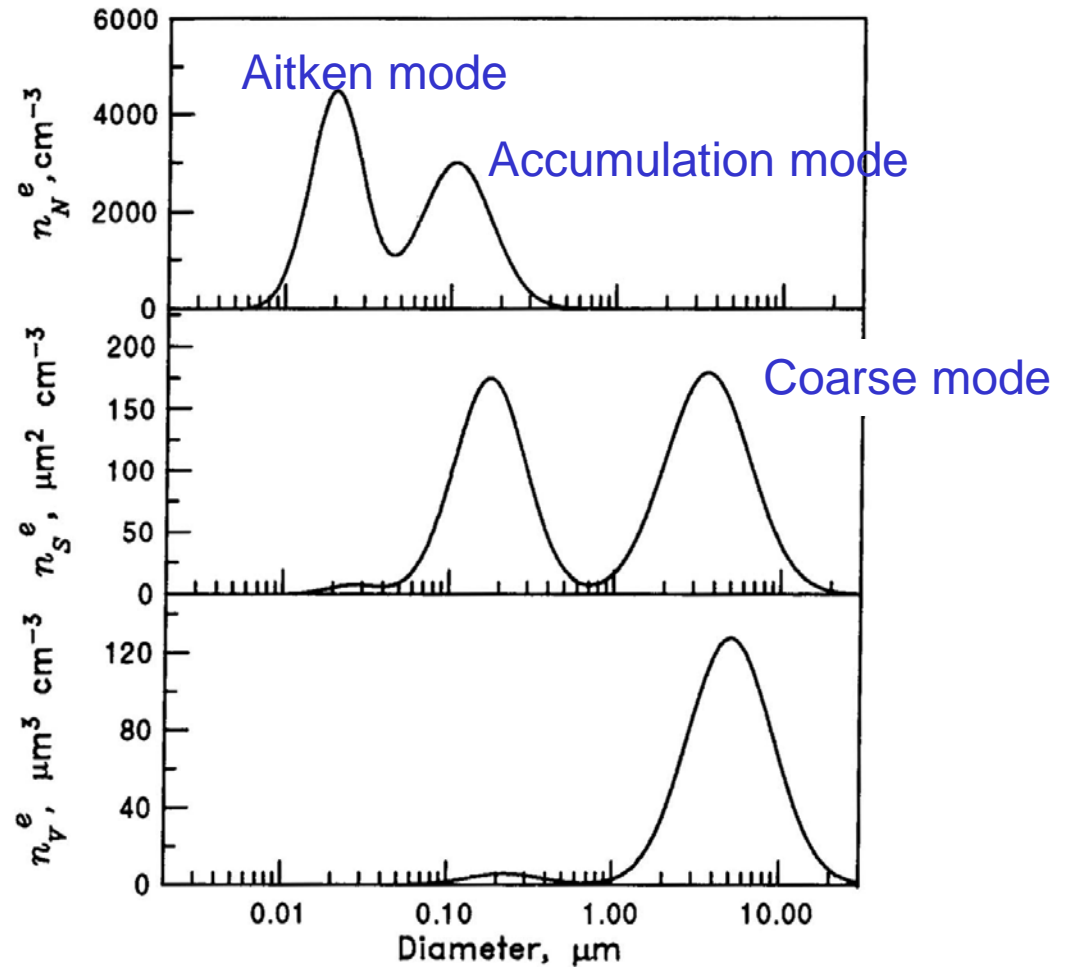
- Total mass concentration is between 1-100  $\mu\text{g m}^{-3}$  (Arctic:  $< 1 \mu\text{g m}^{-3}$ )
  - Total mass of air  $\sim 1 \text{ kg m}^{-3}$
  - $\sim 1\text{-}100 \text{ ppb}_{\text{mass}}$
- Total number concentration is between 100-100,000  $\# \text{ cm}^{-3}$  (Arctic:  $\sim 1\text{-}200 \text{ cm}^{-3}$ )

# Size distributions

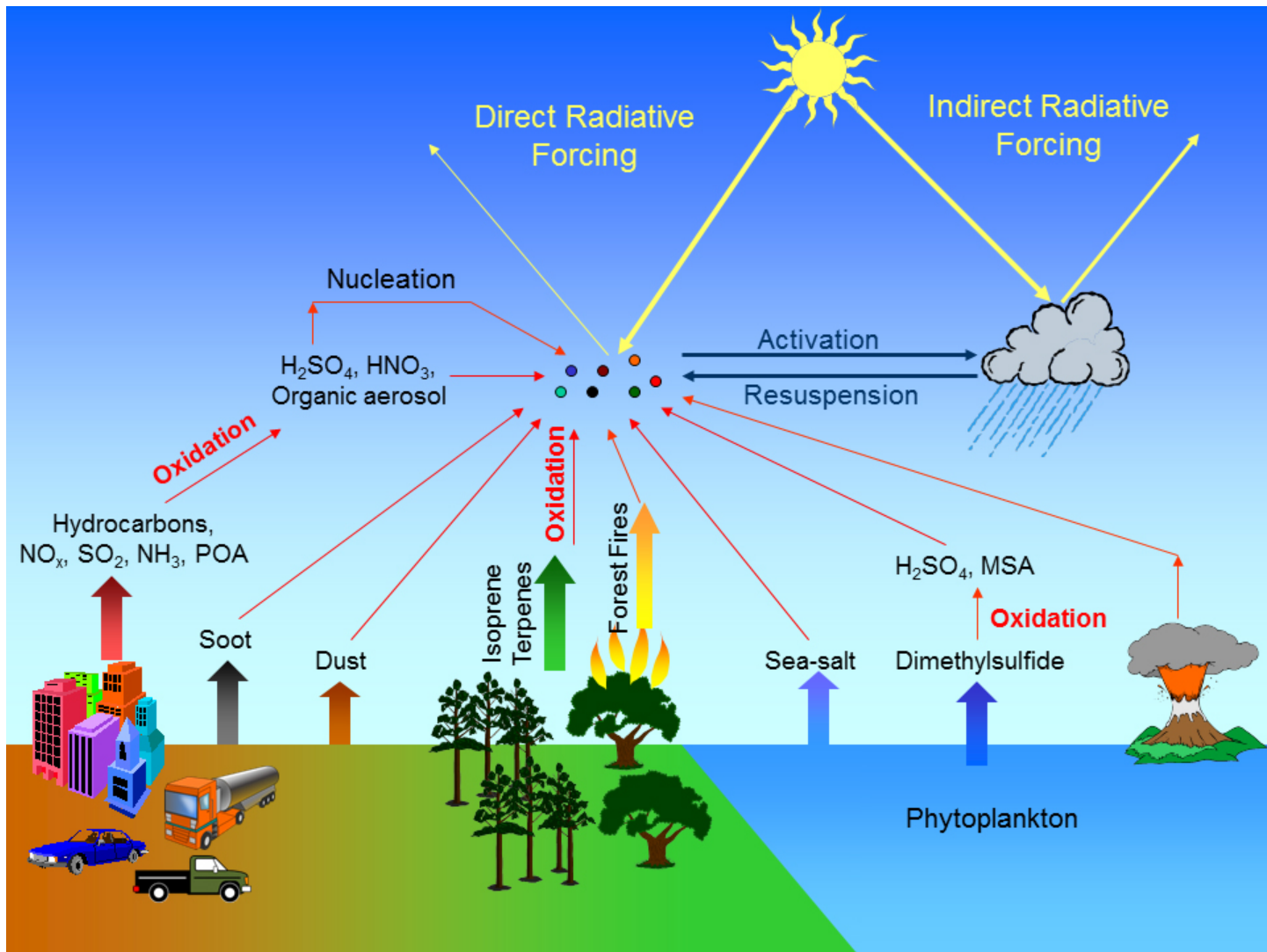
Number distribution  
 $n_n(\log D_p) = dN/d \log D_p$

Surface area distribution  
 $n_s(\log D_p) = dS/d \log D_p$

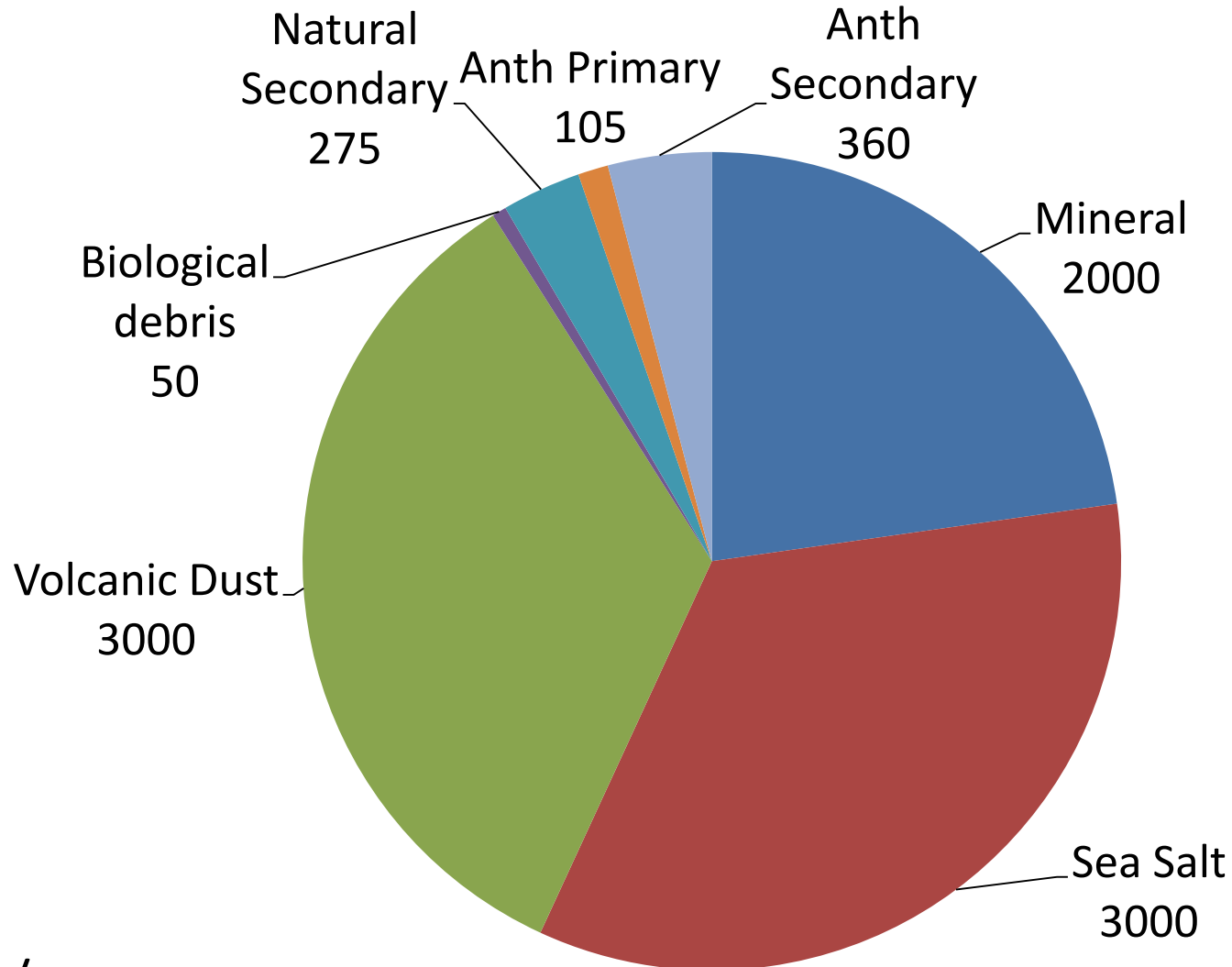
Volume distribution  
 $n_v(\log D_p) = dV/d \log D_p$



# Atmospheric aerosol sources



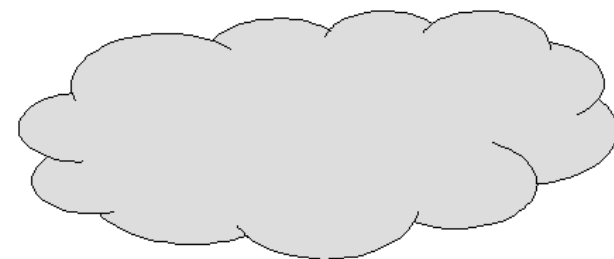
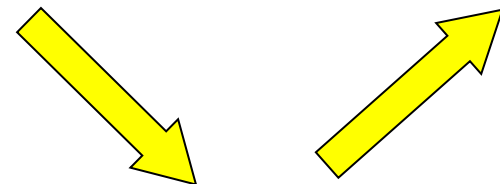
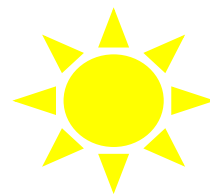
# Global Aerosol Source Contributions



Units: Tg/yr

Seinfeld and Pandis, Atmospheric Chemistry and Physics

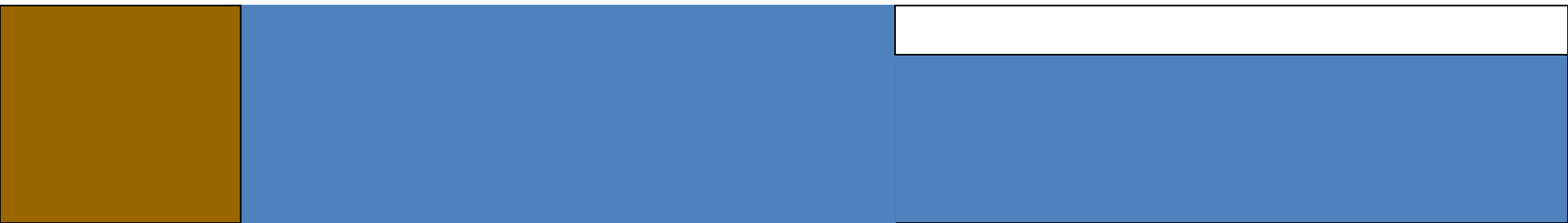
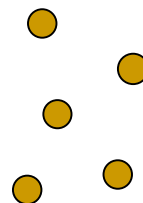
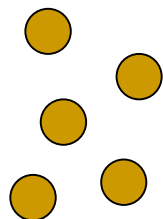
# Aerosols in the Arctic



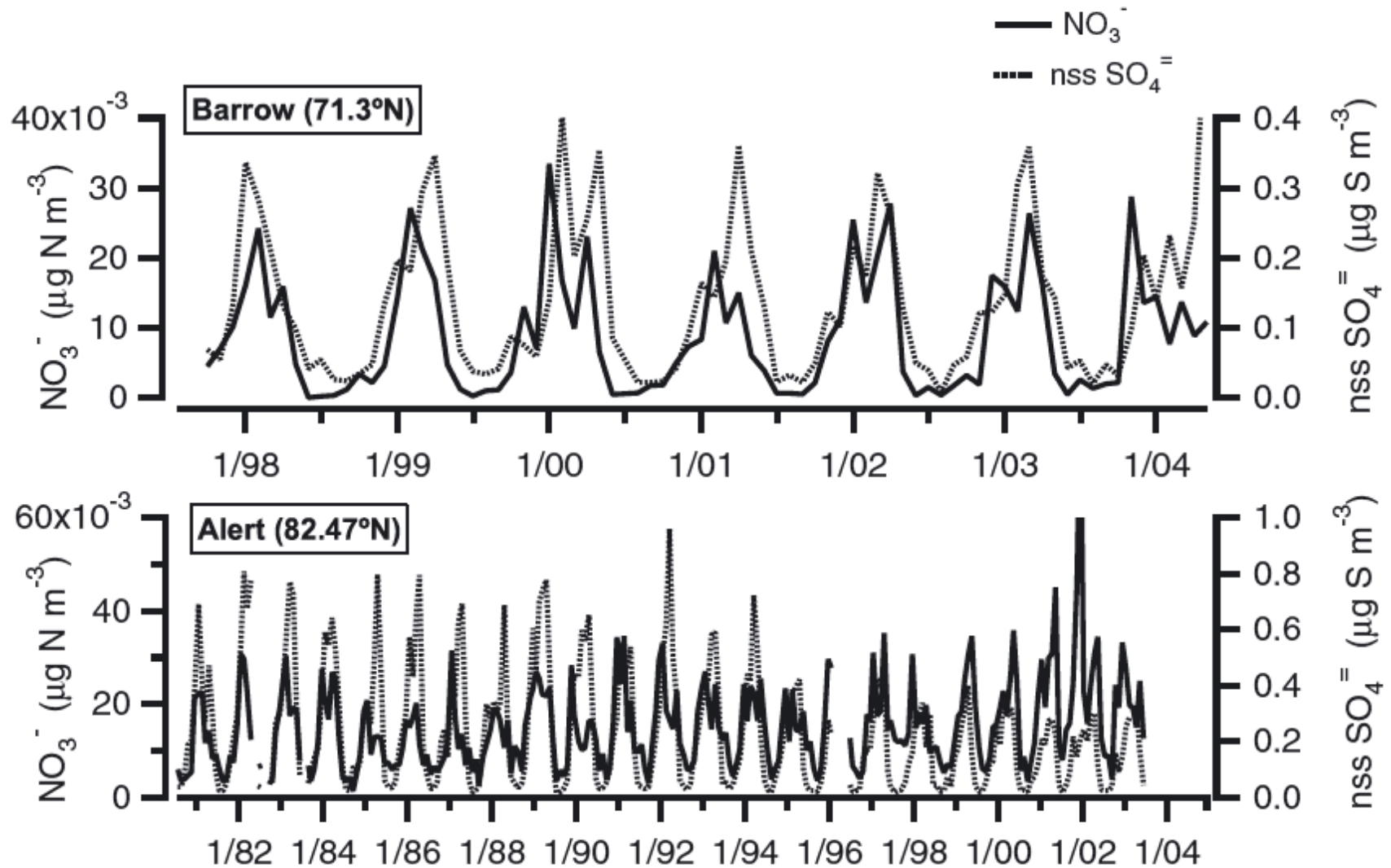
Transport

Sources

Radiation



# Seasonality of Arctic Aerosol Mass

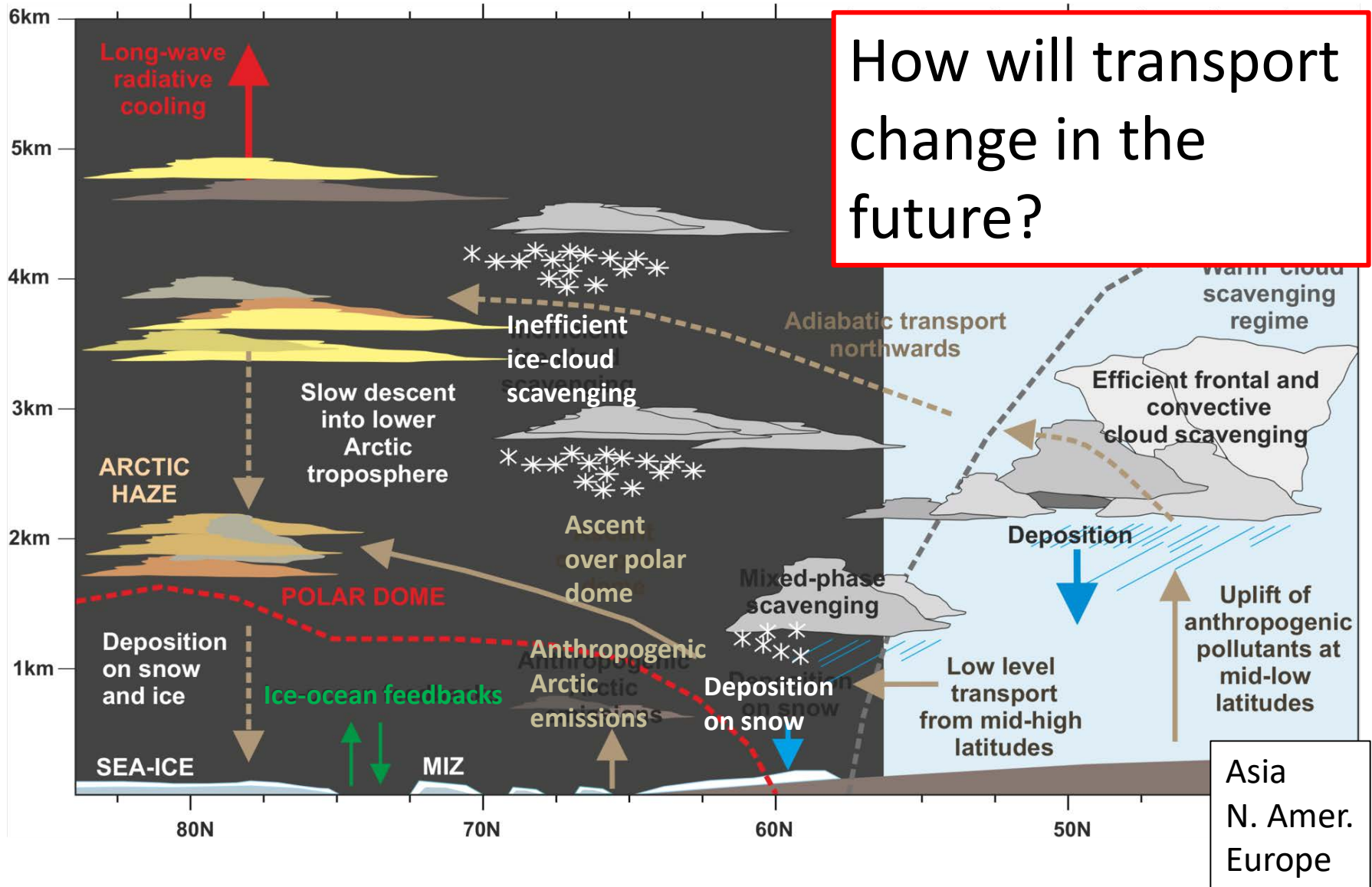


# Arctic Haze



[http://www.esrl.noaa.gov/csd/projects/arcpac/photos/inter\\_dc8.php?num=0](http://www.esrl.noaa.gov/csd/projects/arcpac/photos/inter_dc8.php?num=0)

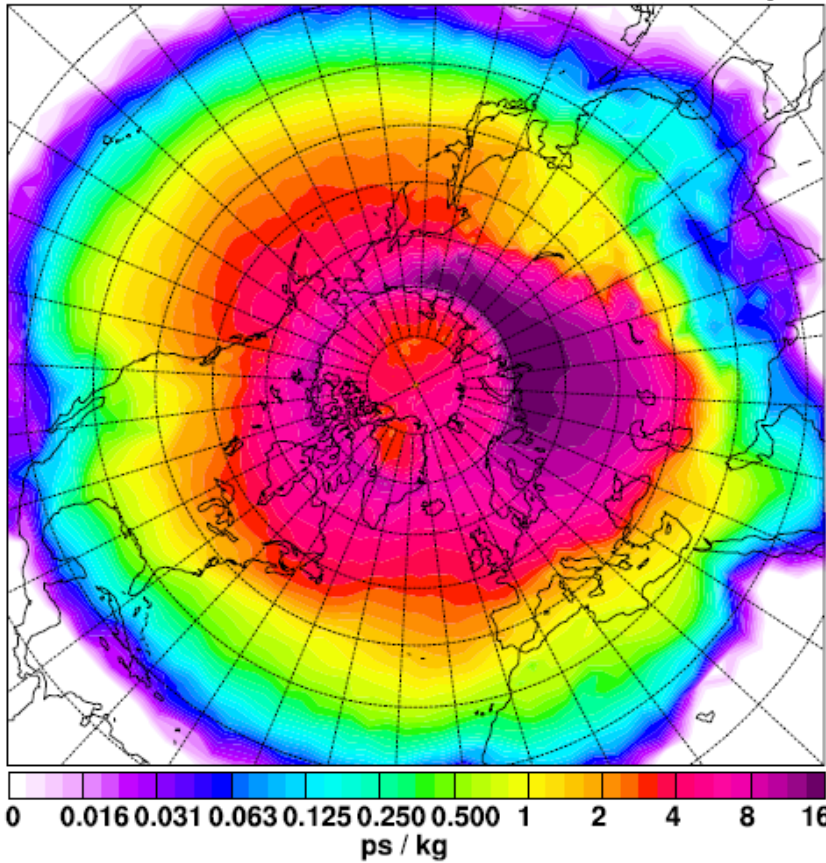
# Winter Transport to Arctic



How will transport change in the future?

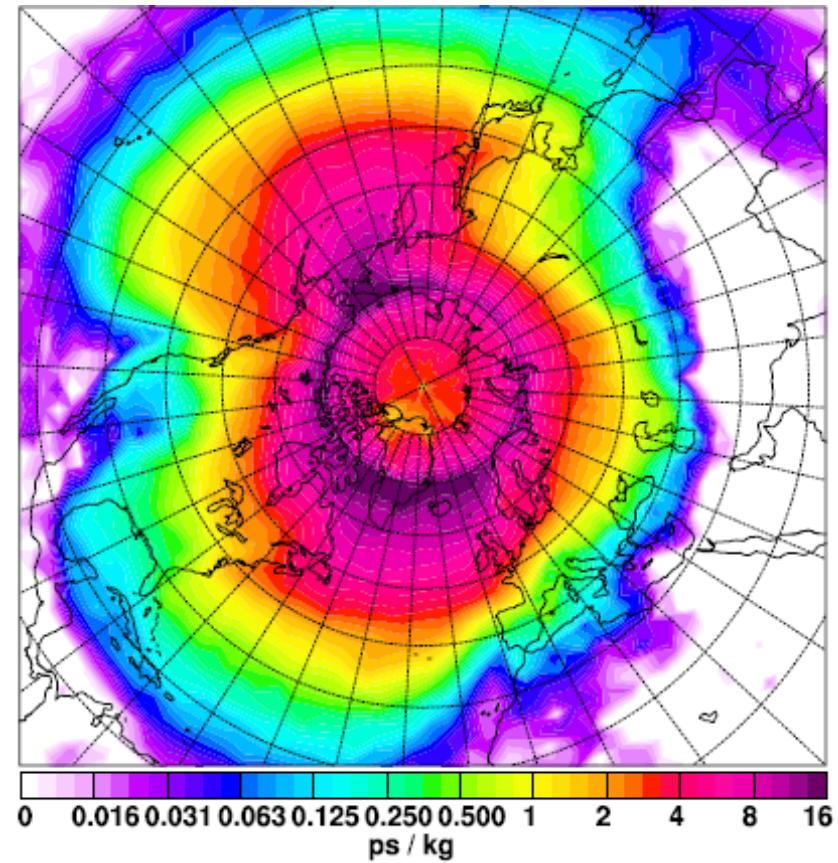
Asia  
N. Amer.  
Europe

# Seasonal Transport into the Arctic



## Winter

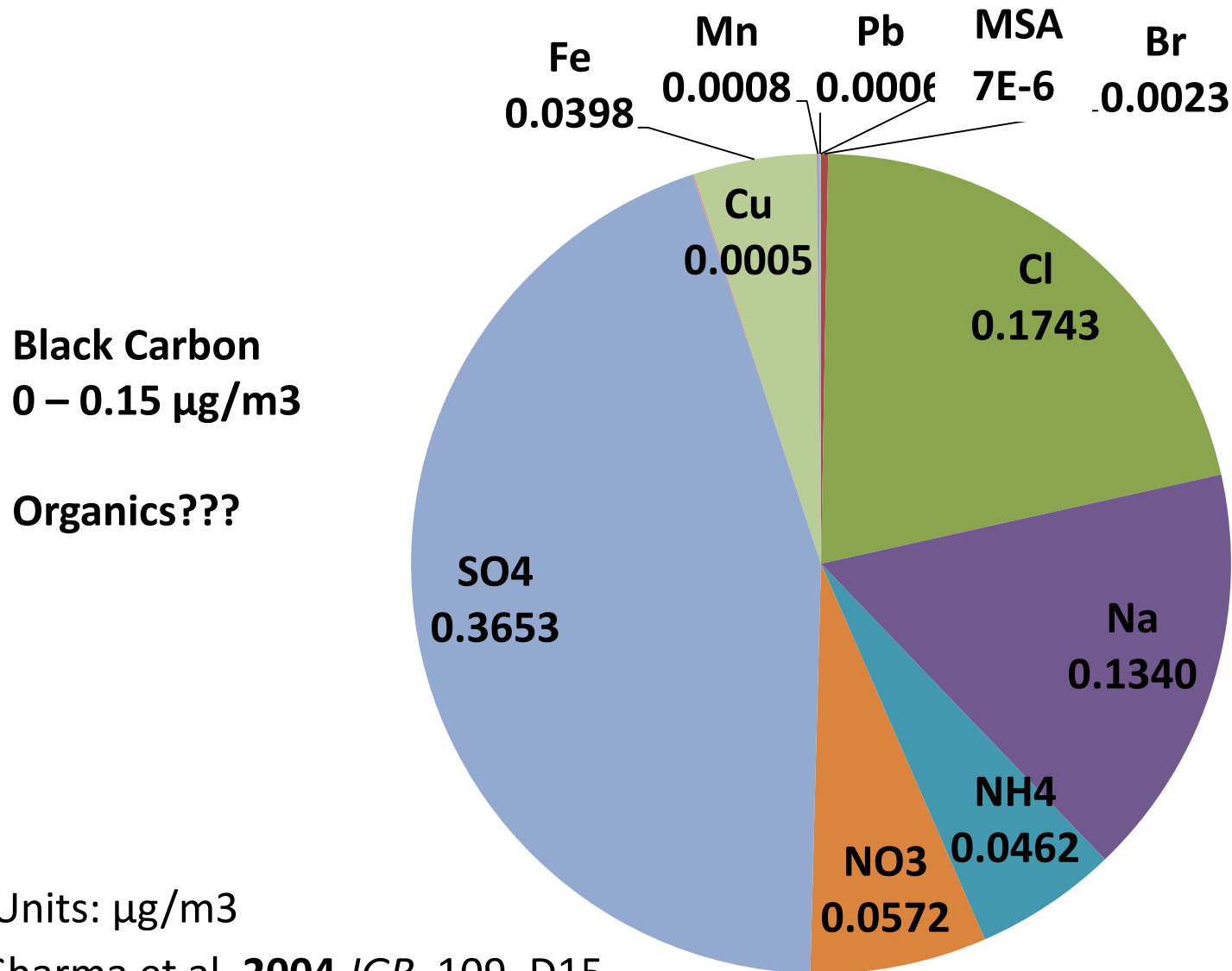
- northern Eurasia influence
- no solar radiation
- stable inversions
- minimal deposition
- shorter transport time



## Summer

- less continental influence
- solar radiation
- convection
- large rate of wet deposition
- longer transport time

# Average Aerosol Composition at Alert: 2006



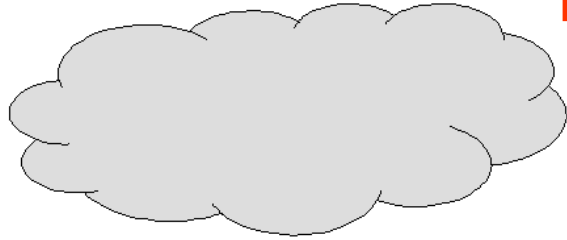
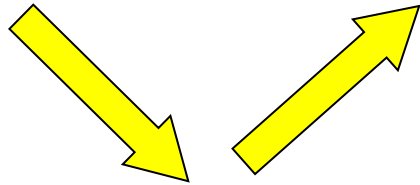
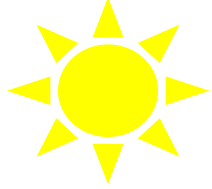
Sharma et al. 2004 *JGR*, 109, D15

<http://www.ec.gc.ca/donneesnatchem-natchemdata/>, accessed Jul 13, 2017

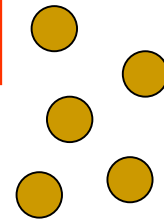
# Sources of Arctic Aerosol: 1980-1995

- Photochemical sulphate
- Anthropogenic sulphate
- Mixed photochemical sulphate / sea salt
- Biogenic sulphur (MSA)
- Sea salt
- Smelter
- Soil
- Nitrate, bromine, iodine
- *Biomass burning*

**Future Sources:  
Shipping  
Oil extraction  
????**



Condensation and Growth



Nucleation and Growth

# Marine Sulphur Chemistry as a Source of Arctic Aerosol



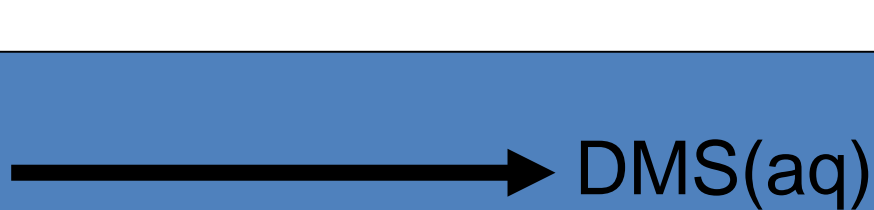
MSA, SO<sub>2</sub>

Oxidation

H<sub>2</sub>SO<sub>4</sub>(g)

Oxidation

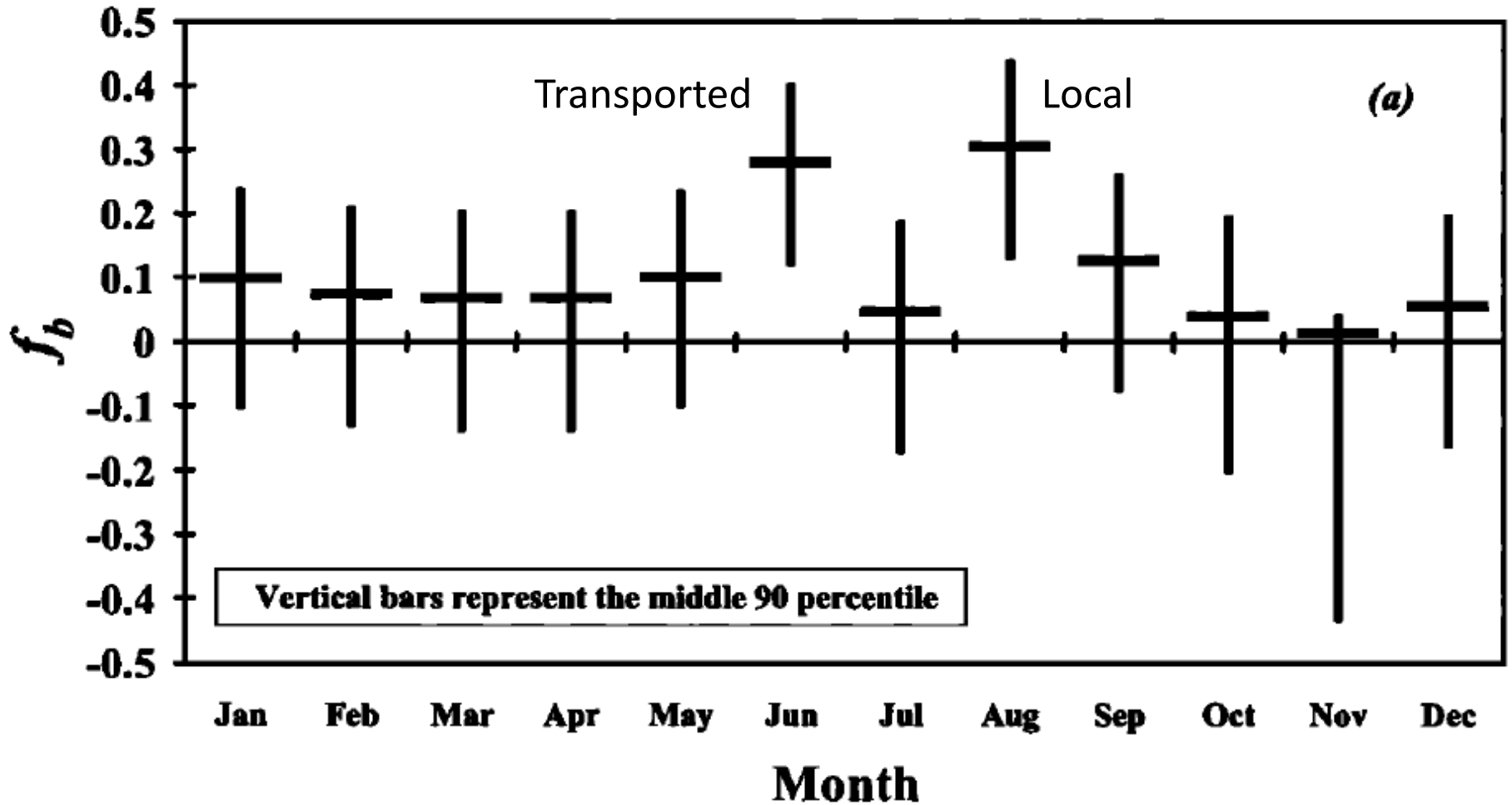
DMS(g)



Phytoplankton

Charlson et al. 1987 *Nature*, 326, 655

# Biogenic sulphate

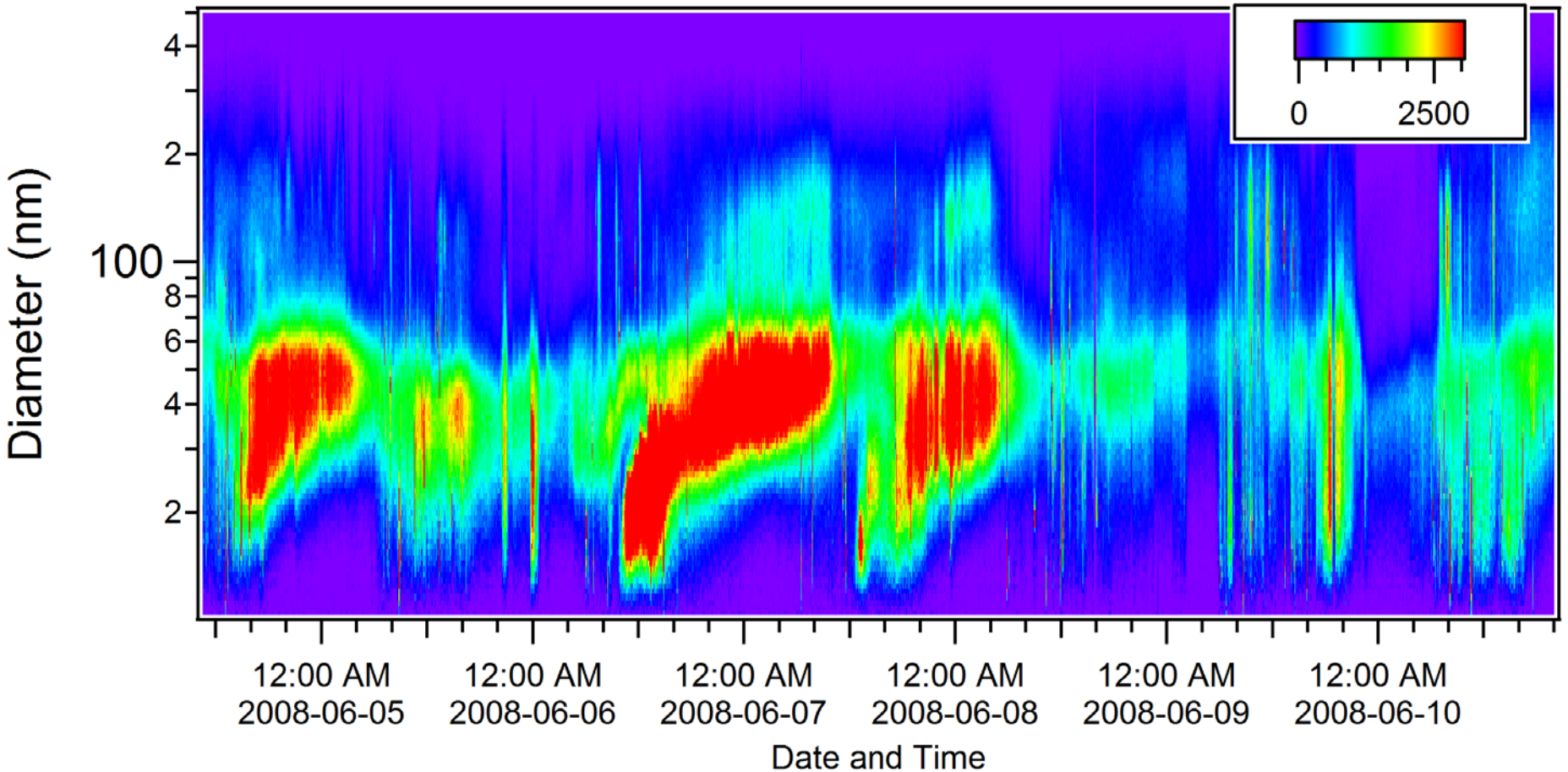


# Nucleation in Marine Environments

- too many existing particles for nucleation to occur in boundary layer
- driven by sulphuric acid
- nucleation in free troposphere
  - colder, cleaner, greater solar radiation

# Nucleation and Growth

# Nucleation / Growth at Whistler Mountain “Banana Plot”



Unpublished data

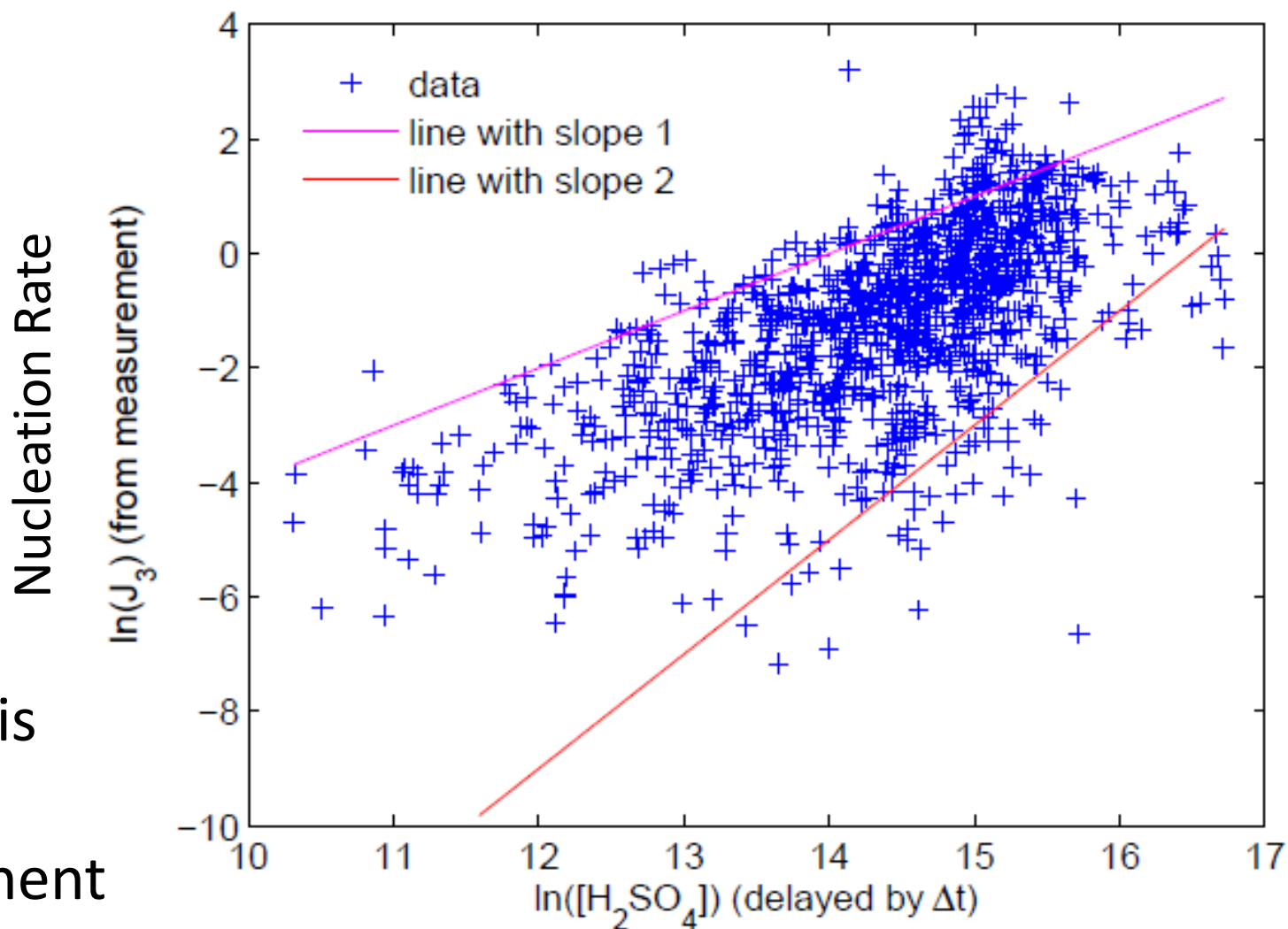
# Ambient Nucleation Rates in Boreal Forest

**Activation**

A:  $0.4\text{-}3.0 \times 10^{-6}$  /s

**Kinetic**

K:  $0.2\text{-}1.4 \times 10^{-12}$  cm<sup>3</sup>/s



H<sub>2</sub>SO<sub>4</sub> is  
main  
component

# Organics Contribute to Nucleation

Grey dots:

Ambient data from Hyytiala

Organics formed by  
TMB oxidation

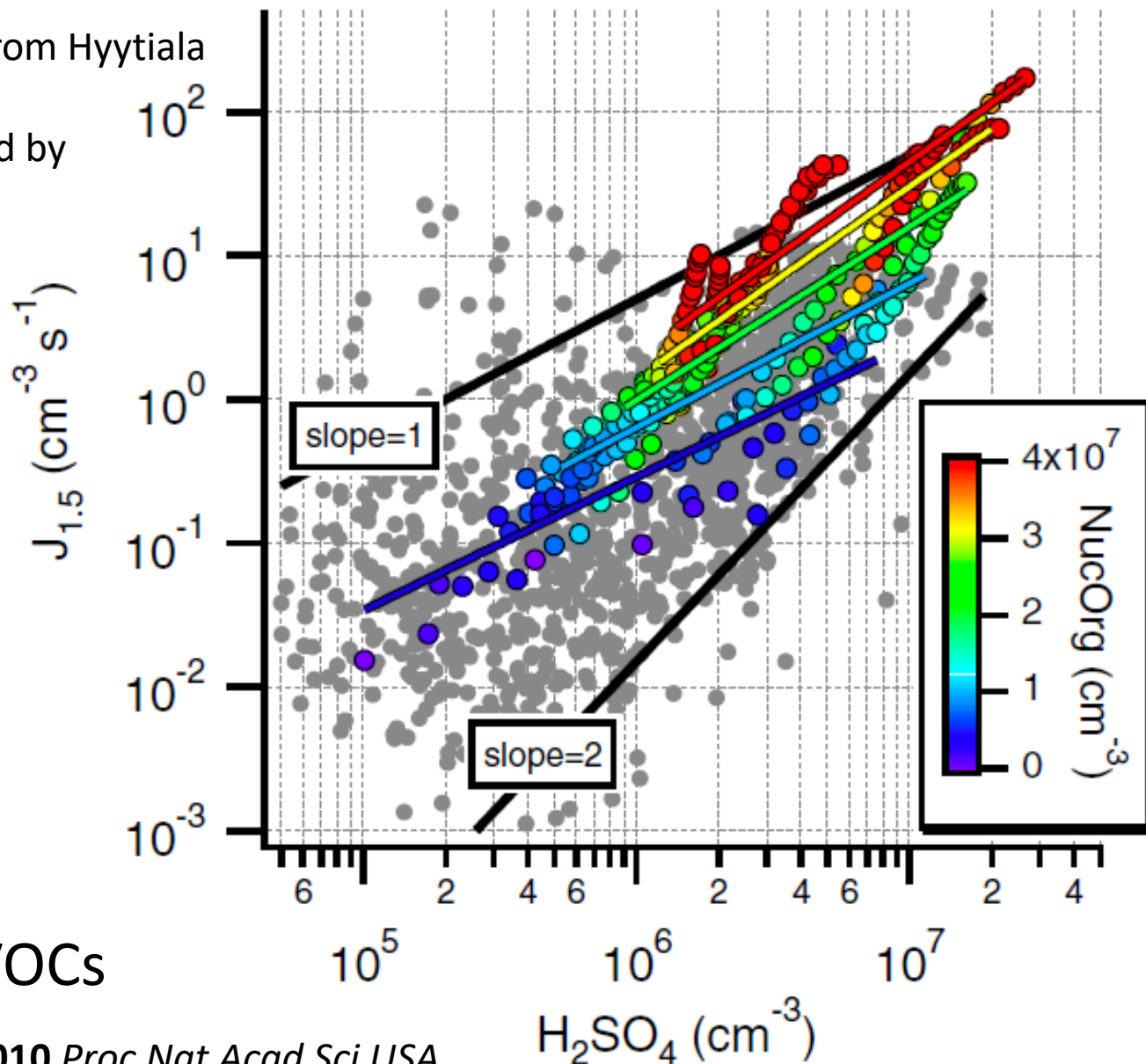
As do:

NH<sub>3</sub>

Amines

Biogenic VOCs

Metzger et al. **2010** *Proc Nat Acad Sci USA*



# Schematic of Particle Formation

## I Small clusters and molecules

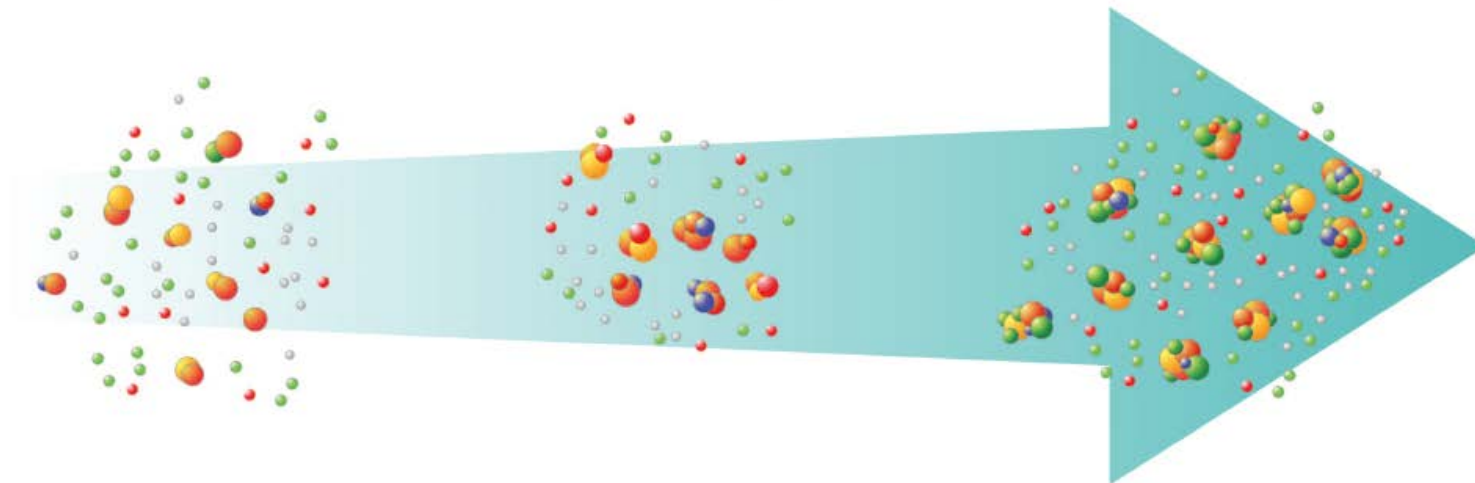
- No direct connection to NPF
- Very slow growth

## II Critical size for clustering

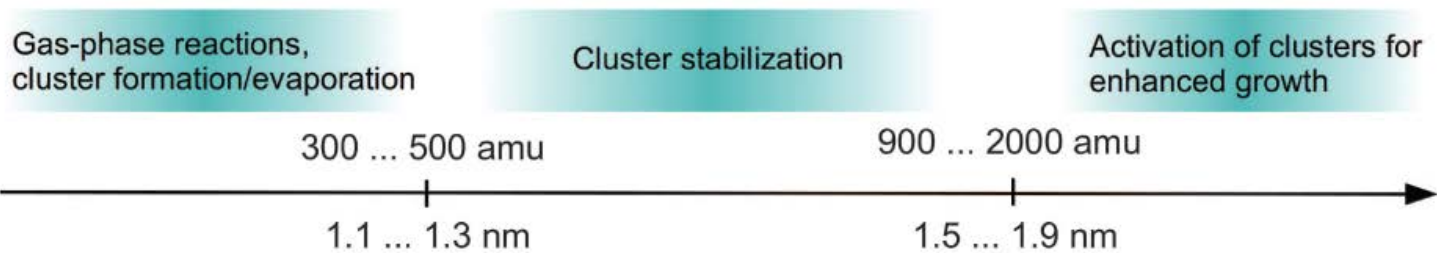
- Sulfuric acid and amines
- Stabilizing organic compounds
- Slowly growing (<1 nm/h)
- Determines  $J_{1.5}$

## III Growing clusters

- Organics start to dominate
- Rapidly growing (~2 nm/h)
- Nano-Köhler
- Determines  $J_3$



### Key processes:



# Nucleation in Marine Environments

- too many existing particles for nucleation to occur in boundary layer
- driven by sulphuric acid
- nucleation in free troposphere
  - colder, cleaner, greater solar radiation

# Particle Growth in Canadian High Arctic

Alert (82.5N, 62.3W)

Eureka  
(80N, 86.5W)



Ridge Lab of the Polar  
Environment Atmospheric  
Research Laboratory (610 m asl)

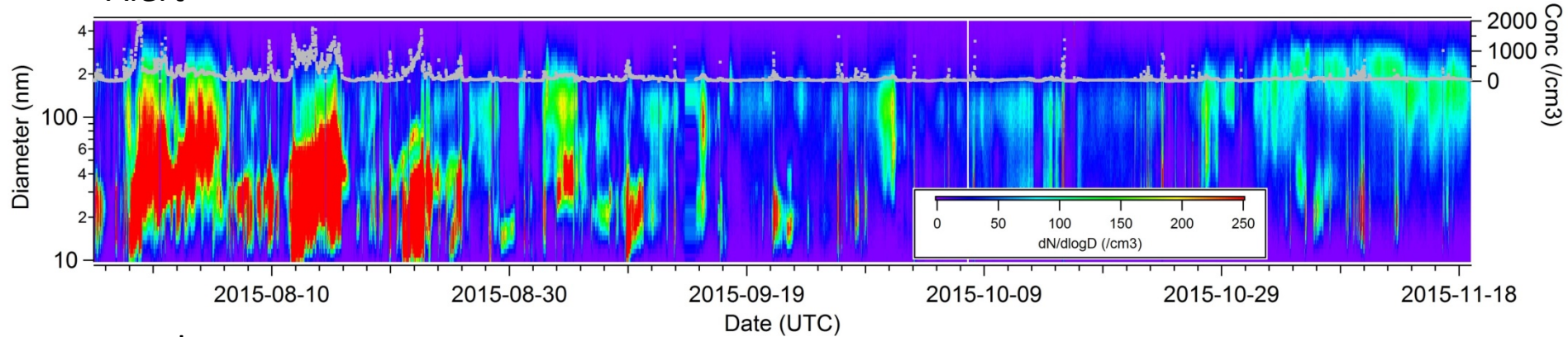


Dr. Neil Trivett Global  
Atmosphere Watch  
Observatory (210 m asl)

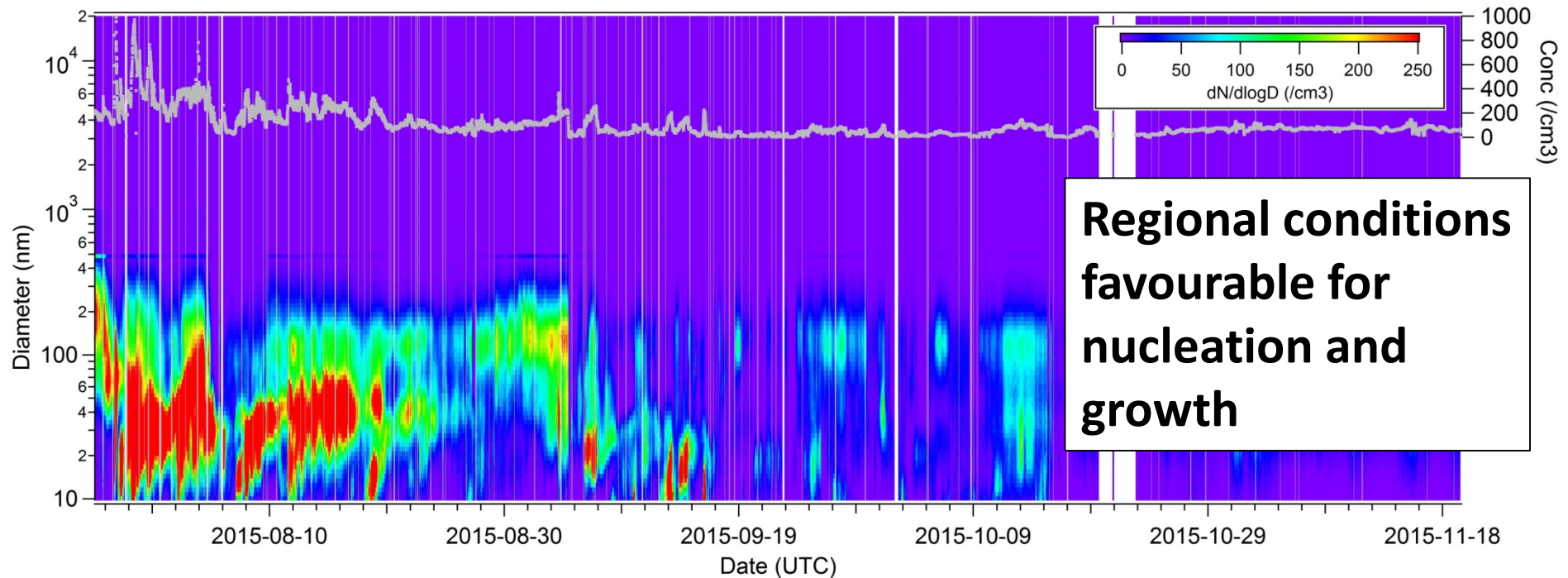
Canada

# Aerosol Size Distribution

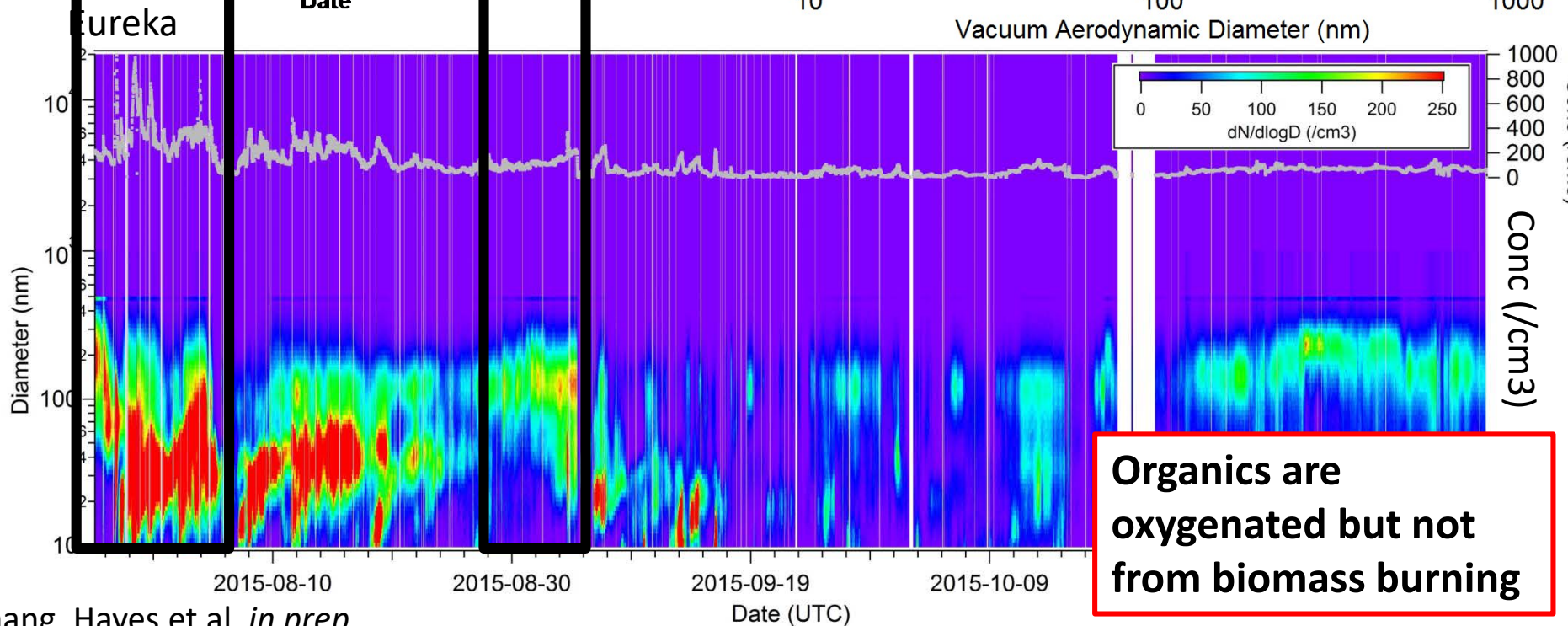
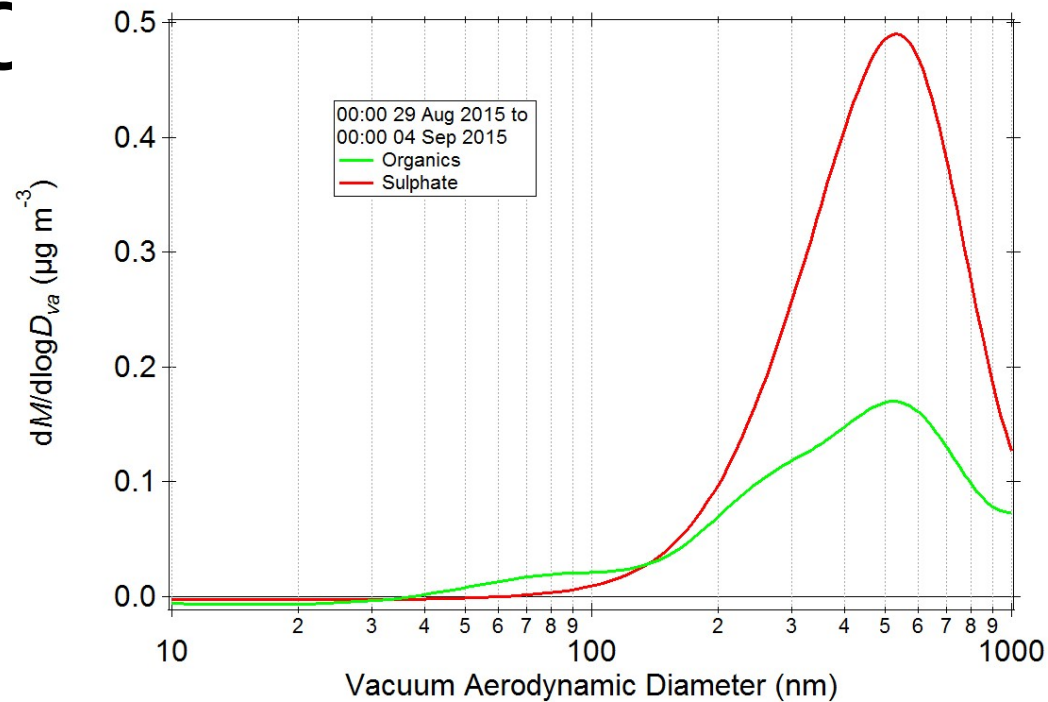
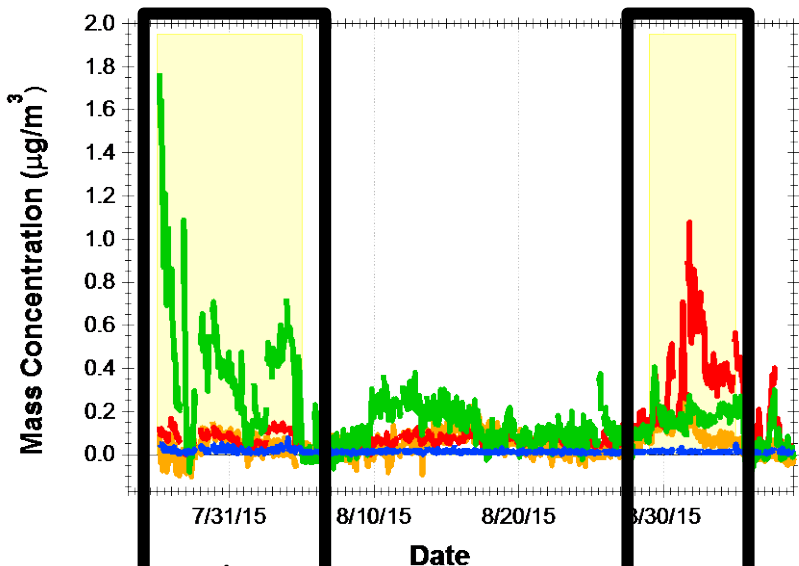
Alert



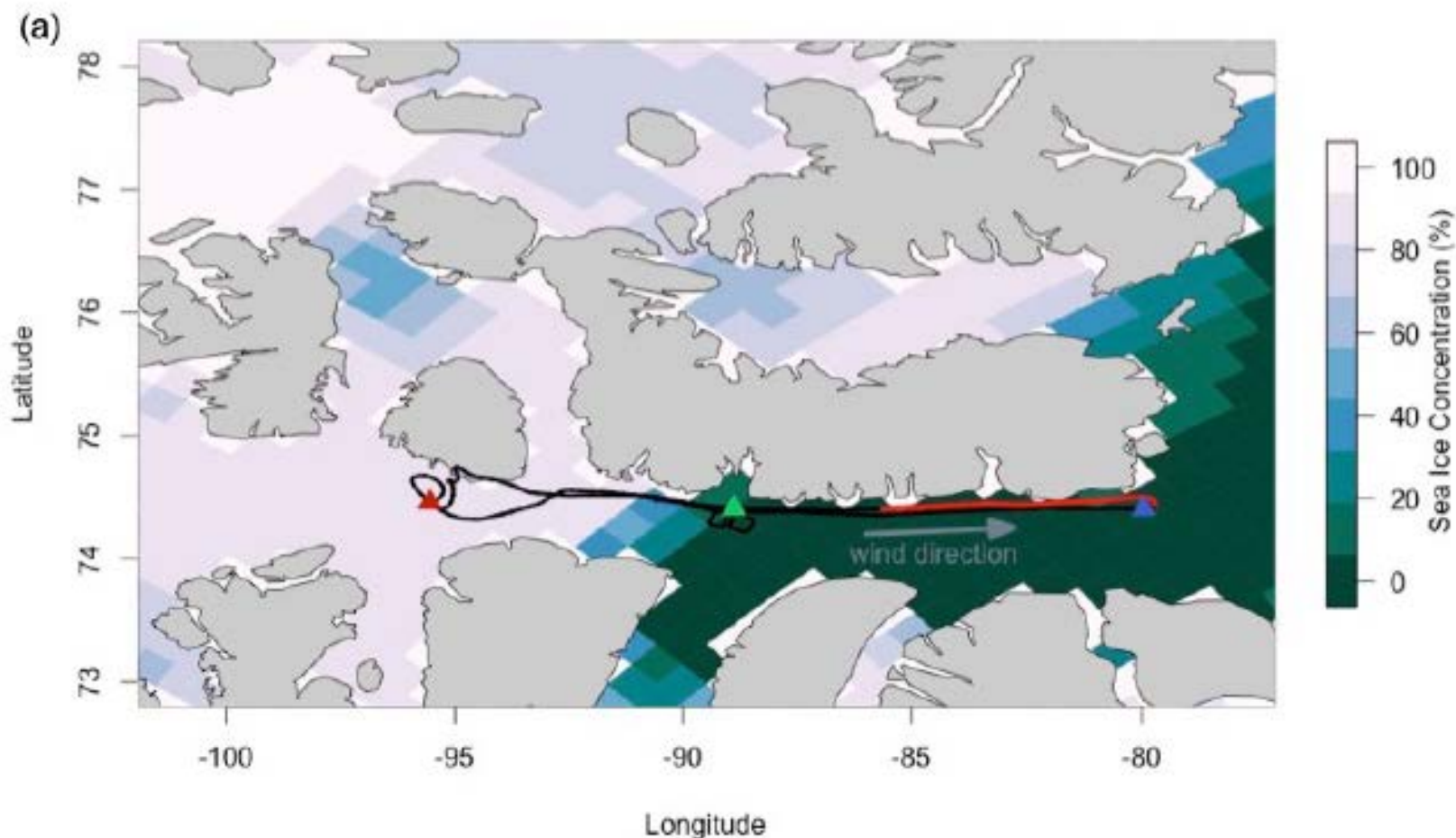
Eureka



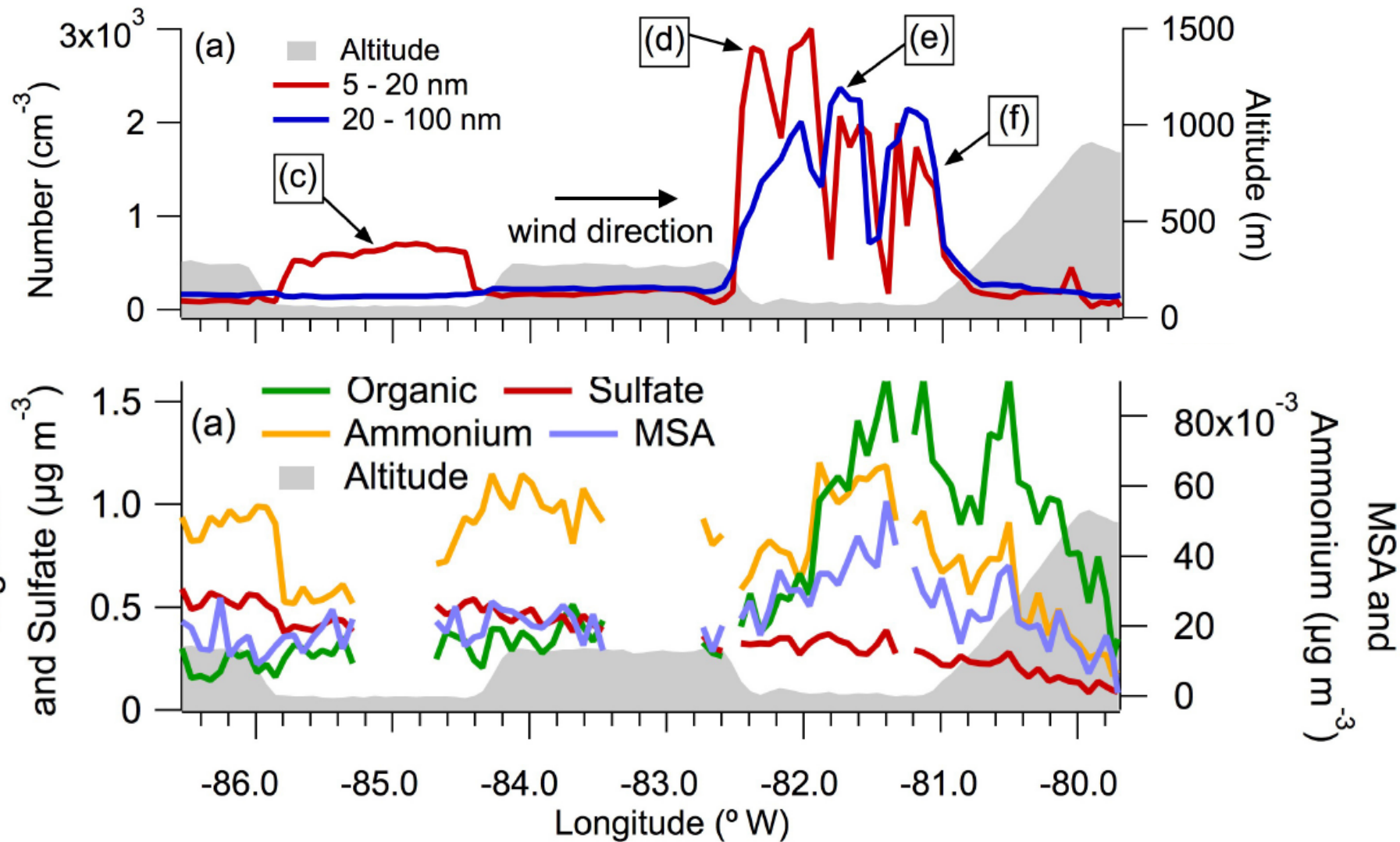
# Eureka Aerosol C



# NETCARE: Marine Emissions and Organics Contribute to Aerosol Growth



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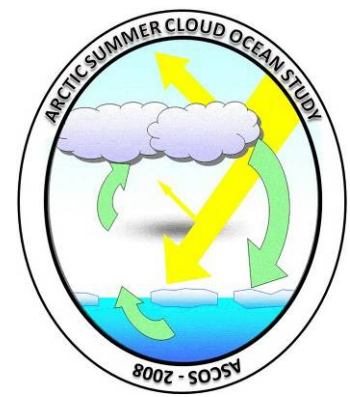
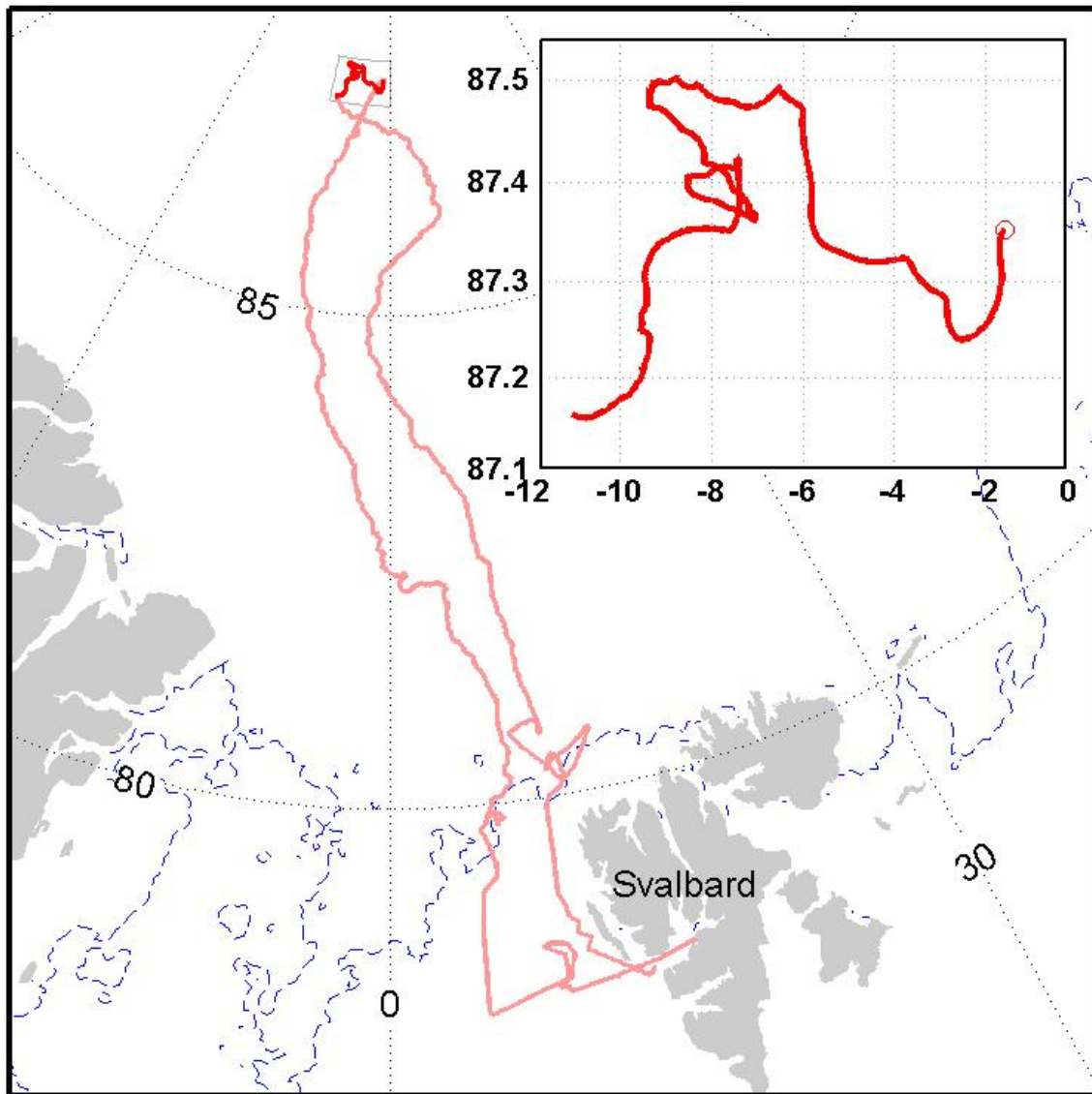
# Other NETCARE Results

- E. Mungall – volatile organics originating from ocean surface
- J. Burkart, D. Collins – particle growth events in Canadian Arctic

# Summer Arctic Aerosol Composition and Sources

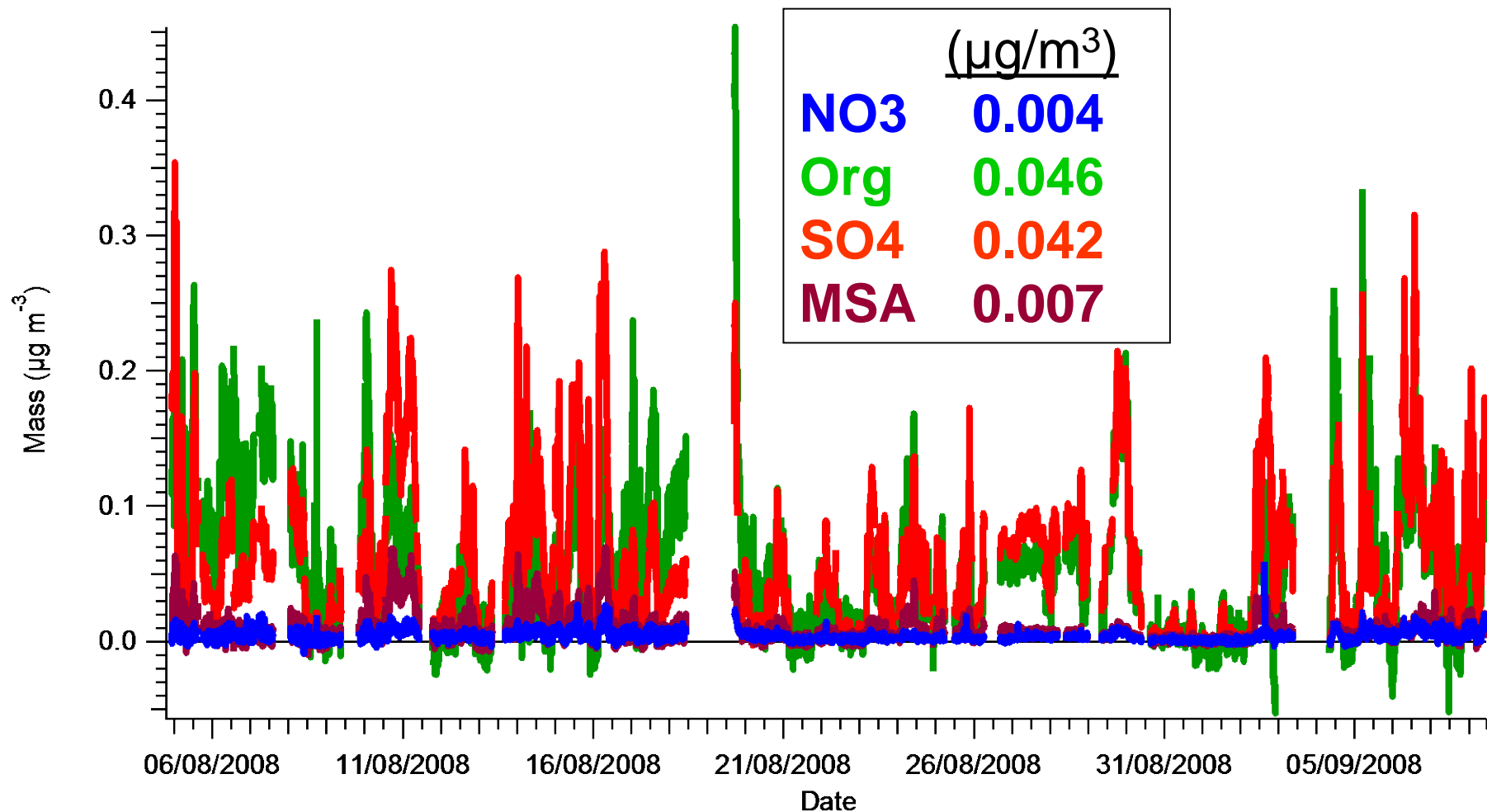
# Arctic Summer Cloud and Ocean Study

## Aug 1 – Sep 9, 2008



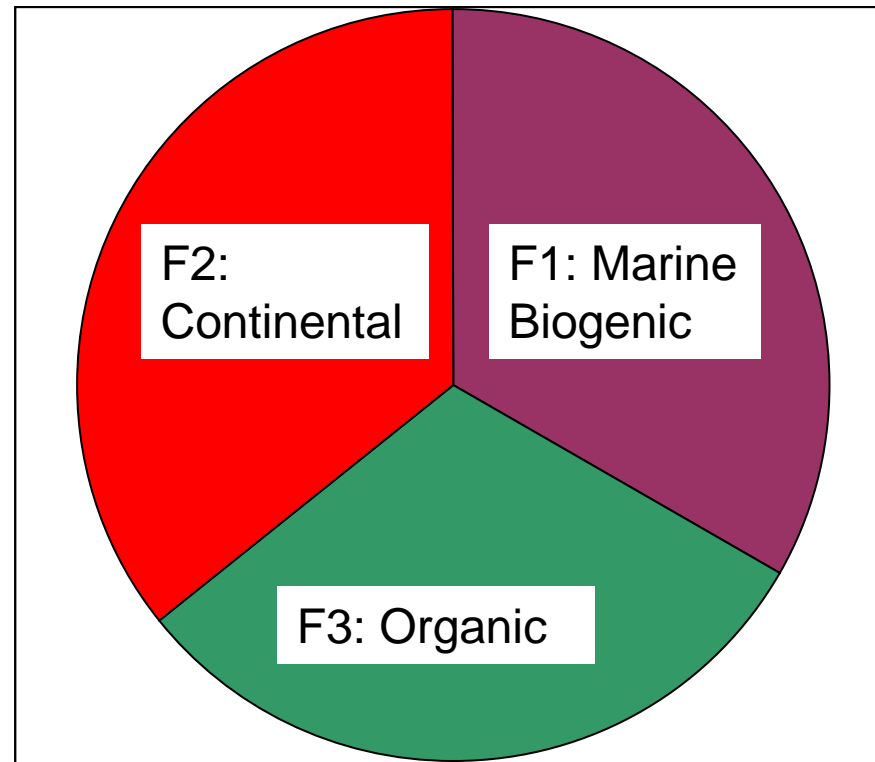
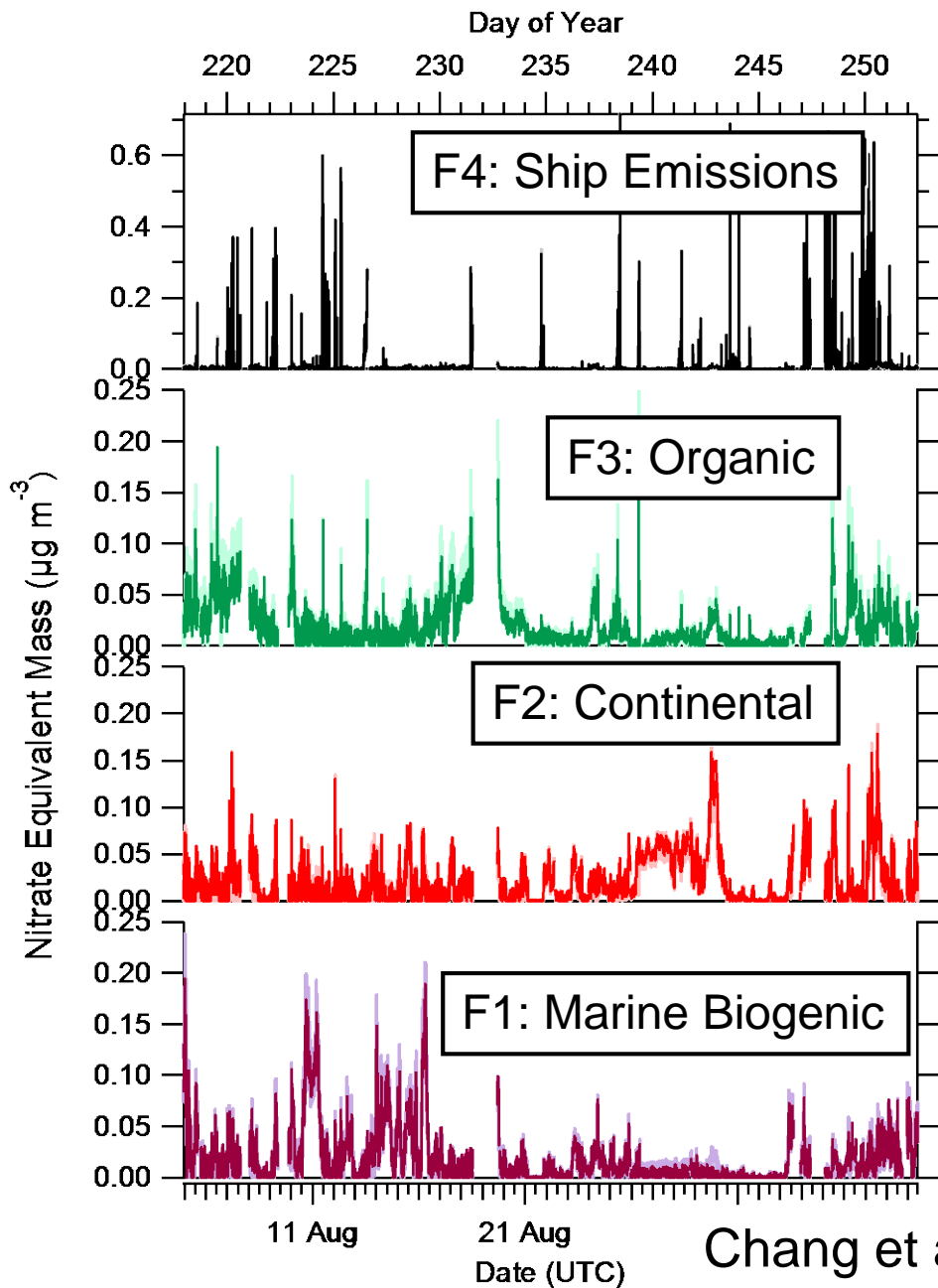
map courtesy of I. Brooks

# Aerosol Chemical Composition Measured by AMS



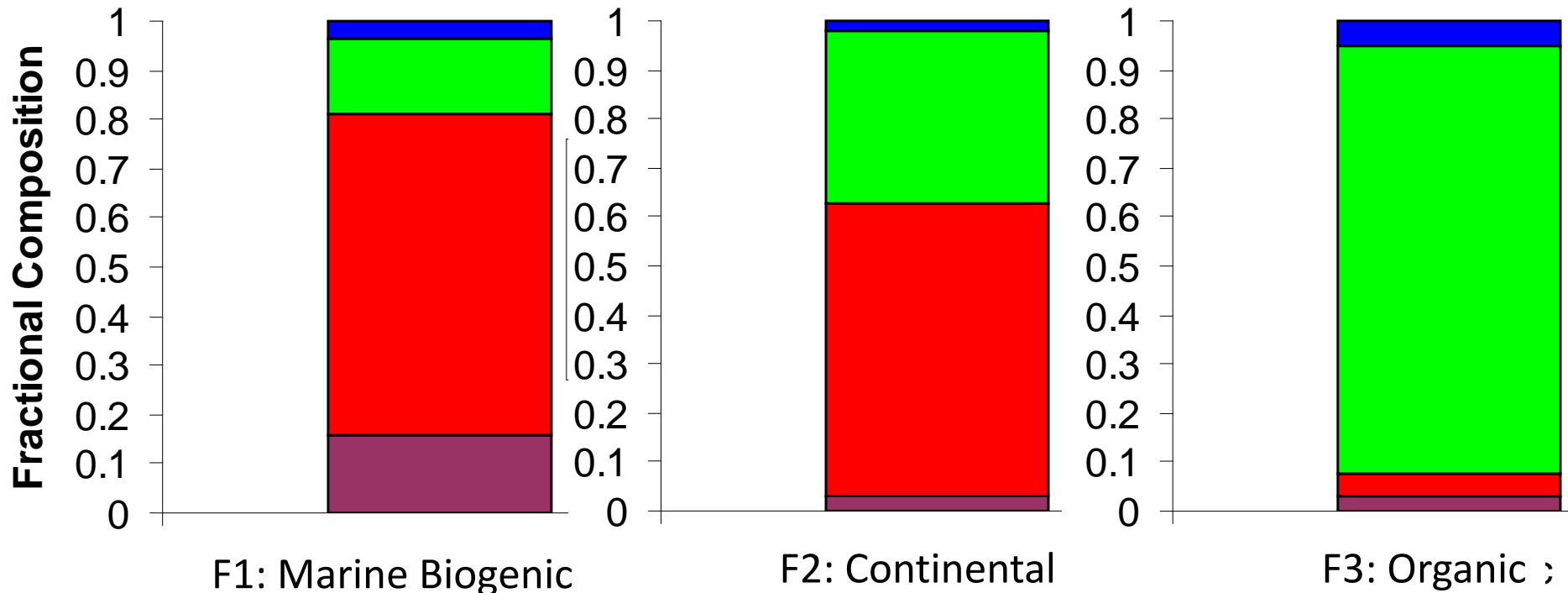
MSA fragmentation from Langley et al. **2010** *Atmos. Chem. Phys.*

# Factor Analysis



# Composition of Factors

NO<sub>3</sub>   Org   SO<sub>4</sub>   MSA



# Summary

- winter aerosol is transported
- summer aerosol is more local
  - organic component is important
  - nucleation and growth observed and linked to marine sources
  - How will this change in future climate?
  - How will industrial activities affect aerosol processes?