

# Arctic Halogen Chemistry

## Part II

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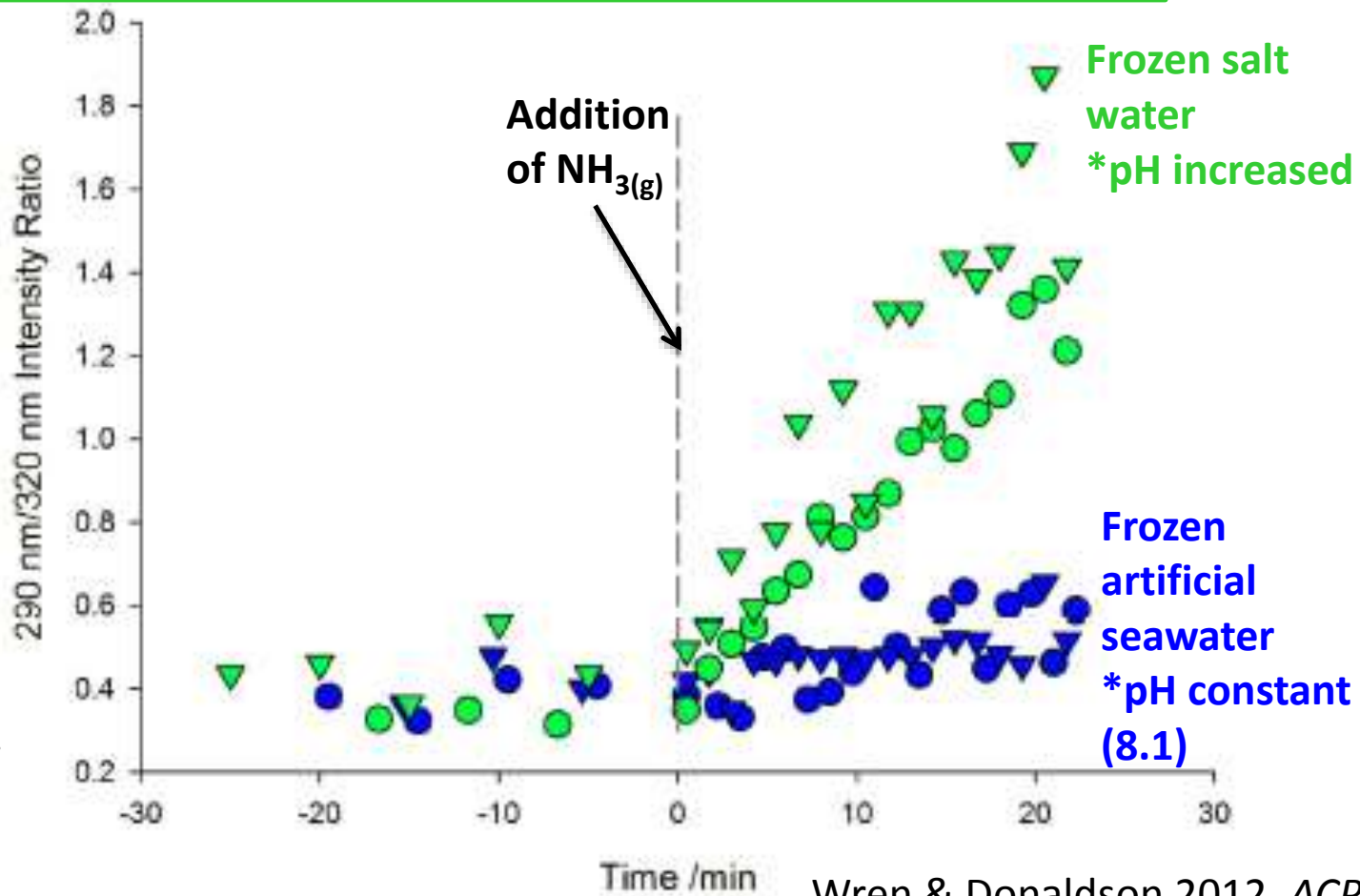




# Sea Ice does not Produce $\text{Br}_2$ (Pratt et al. 2013)

## → Sea Ice is Buffered Against pH Change

Lab Study: Frozen seawater surface buffered against pH change from gas deposition



Glancing-angle laser induced fluorescence with surface-active, pH-sensitive fluorescent dye (harmine)

# Frost Flowers – Br<sub>2</sub> Production Not Expected



Br<sup>-</sup>/Cl<sup>-</sup>: ~1/900 - 1/700 (too low)

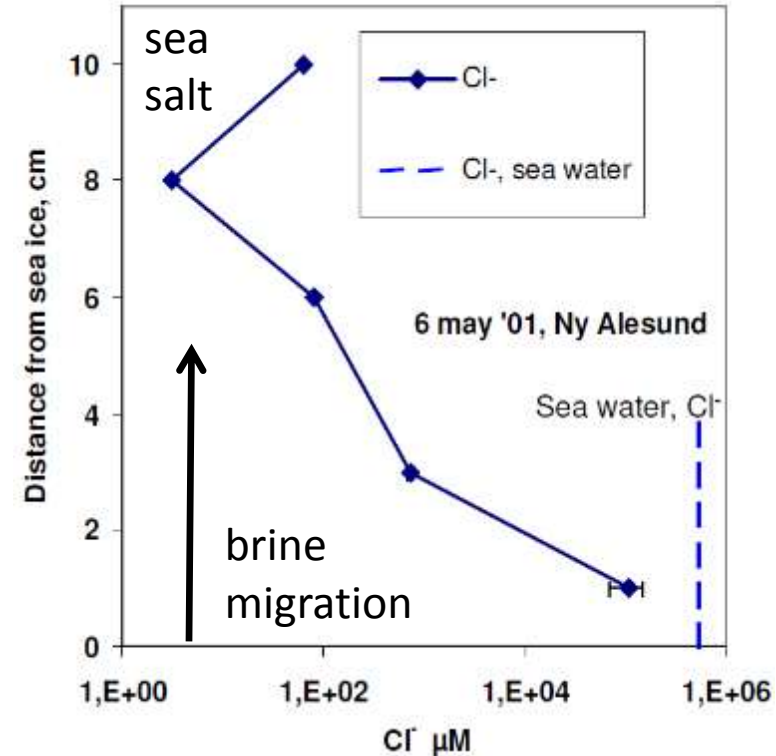
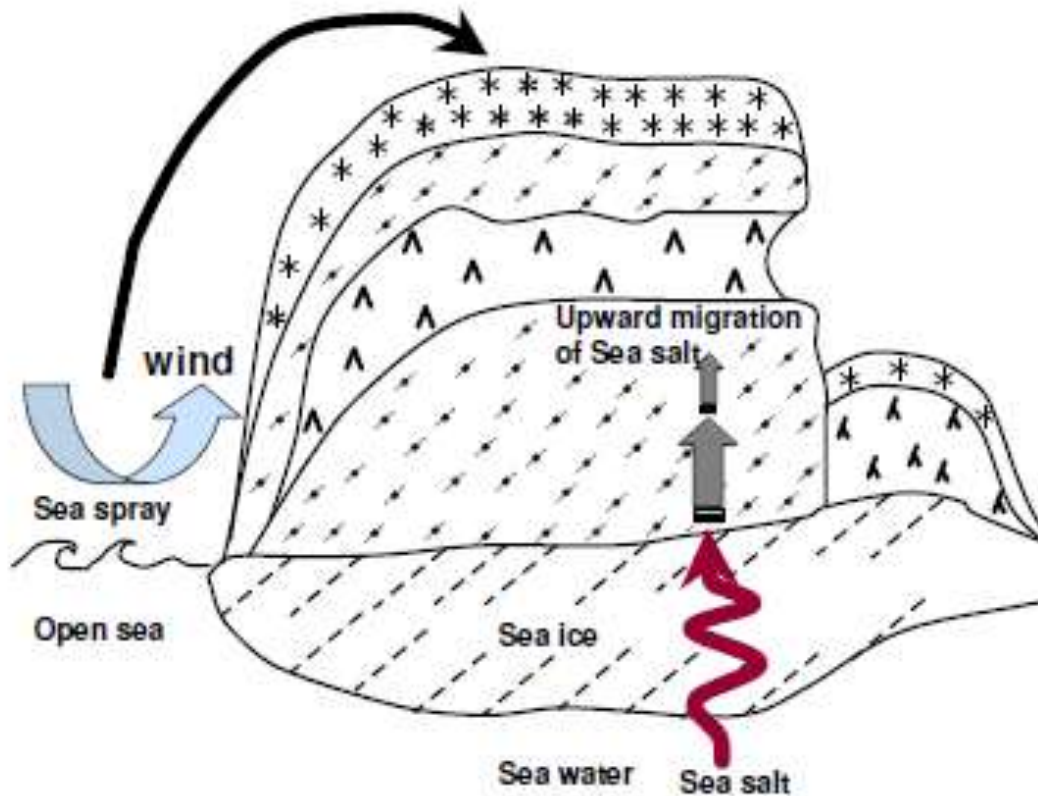
pH: 7.4-8.7 (too high)

(Douglas et al 2012, *JGR*)

“Frost flowers are unlikely to be a direct source of atmospheric bromine [due to pH].” (Kalnajs & Avallone 2006, *GRL*)

Wind + frost flowers does not produce aerosols in the lab (Roscoe et al 2011, *JGR*; Yang et al 2017, *ACP*).

# Sea Spray Deposition to Surface Snowpack



Domine et al. 2004, *ACP*

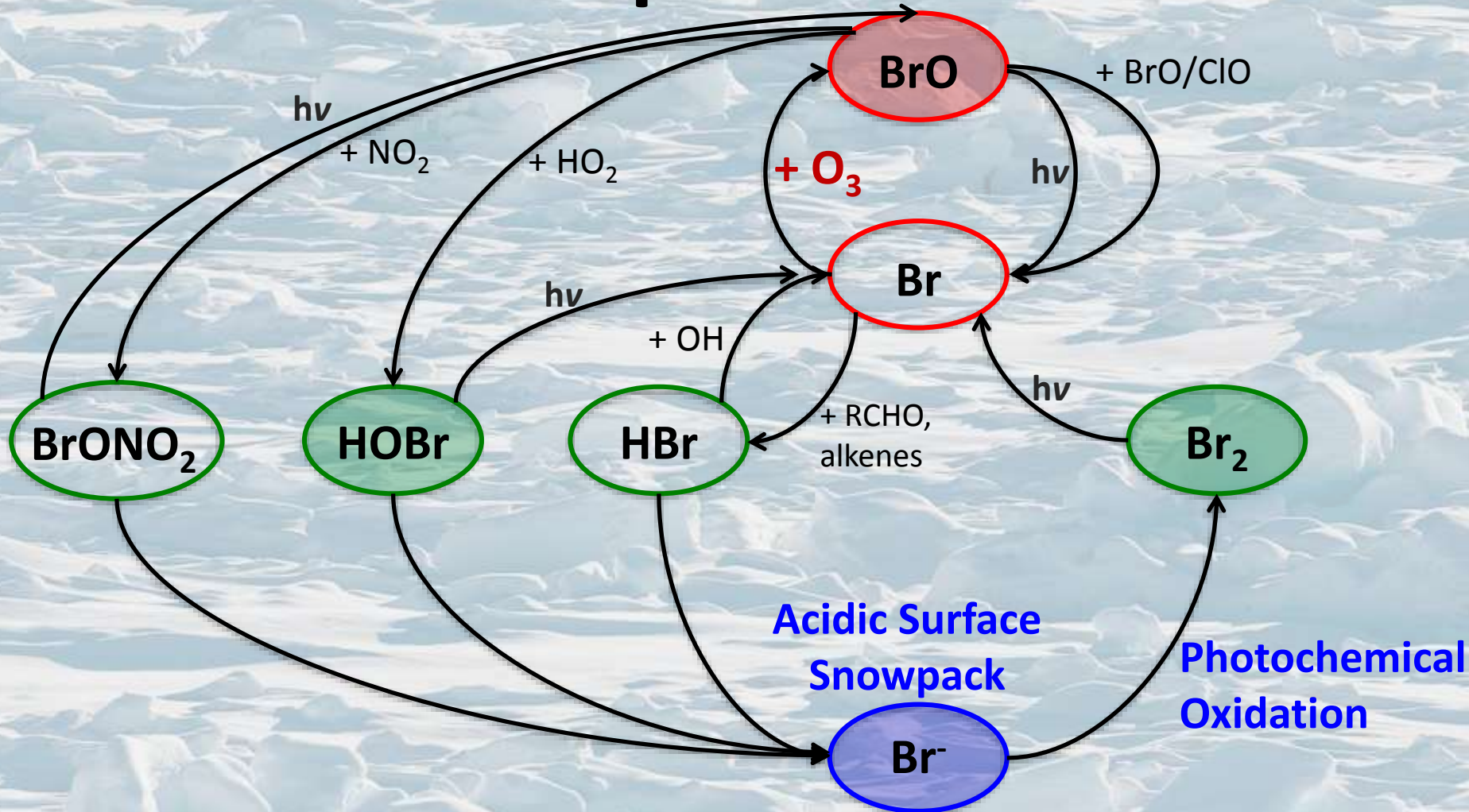
Arctic first-year sea ice covered by ~29 cm (average) of snow in springtime (Kwok et al. 2011, *JGR*)

**Atmospheric deposition determines chemistry of surface snow**

**→ impacts  $[\text{Br}^-]/[\text{Cl}^-]$  & pH**

**→ bromine activation**

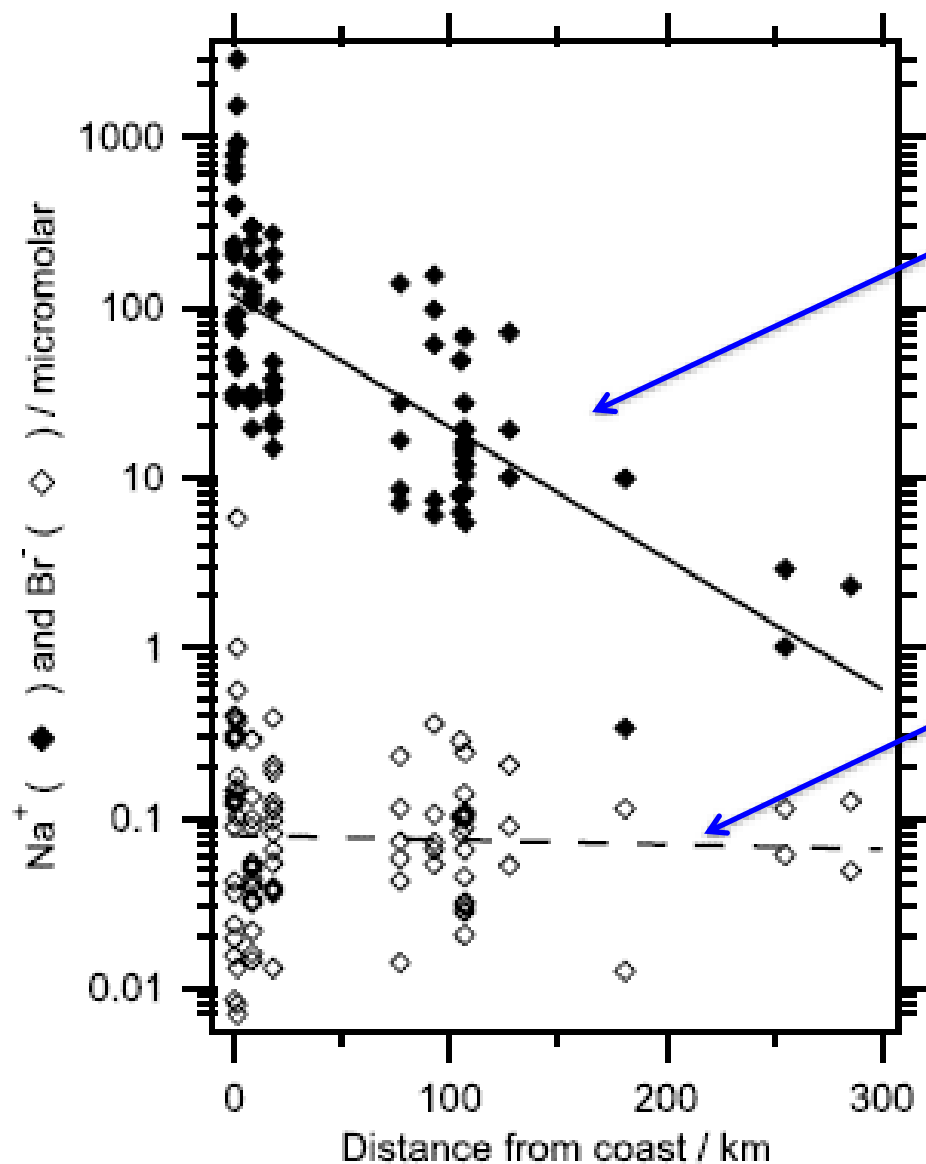
# “Bromine Explosion” → Ozone Depletion Events



Barrie et al. 1988, *Nature*  
Finlayson-Pitts et al. 1990, *Nature*  
Fan & Jacob 1992, *Nature*

McConnell et al. 1992, *Nature*  
Abbatt et al. 2012, *Atmos. Chem. Phys.*  
Pratt et al. 2013, *Nature Geoscience*

# Inland Tundra Snow as Bromine Source

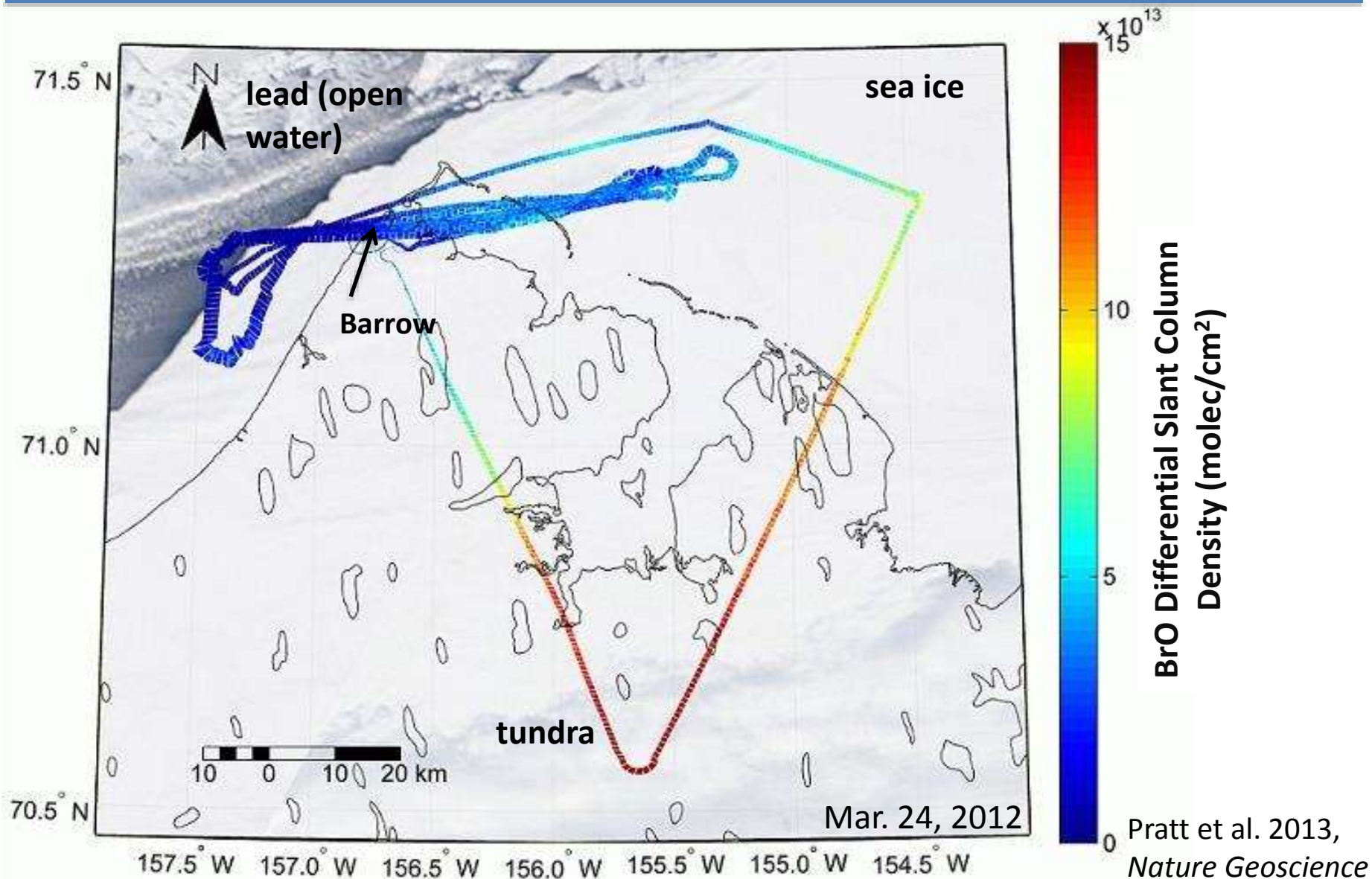


Decreasing  $\text{Na}^+$  inland due to increased deposition of sea spray aerosol

Transport and inland deposition of trace bromine gases (e.g., HOBr, HBr,  $\text{BrNO}_2$ )

- 75-90% of deposited bromine is estimated to be re-emitted as  $\text{Br}_2/\text{BrCl}$

# BrO Enhanced Near Surface over Snow-Covered Tundra



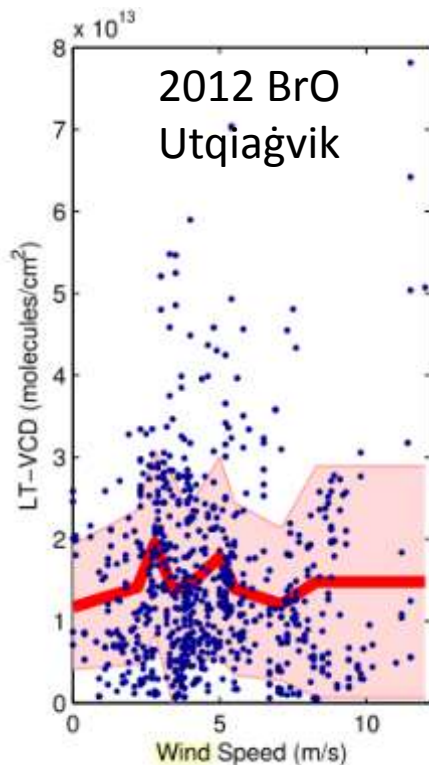
# Additional Arctic Br<sub>2</sub> Sources?

## ? Surface snow above MYI (*likely*)

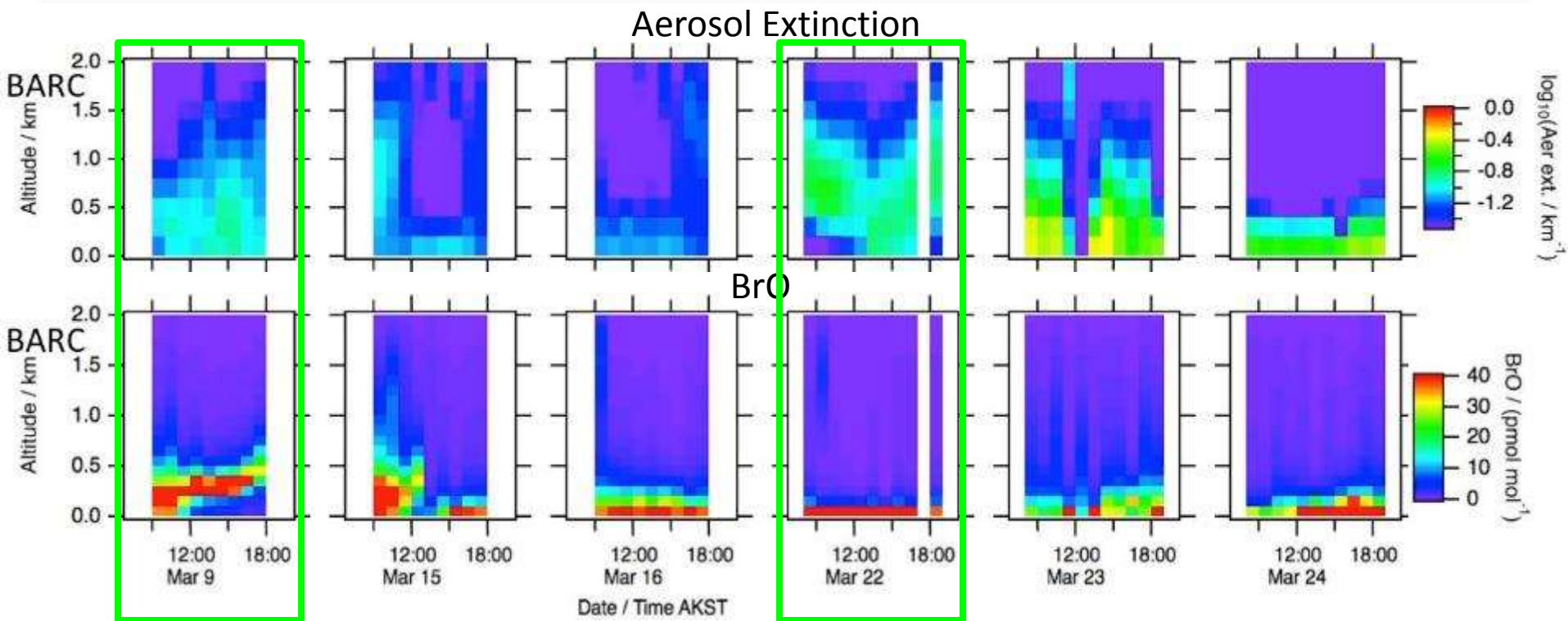
- Similar pH, [Br<sup>-</sup>], and [Cl<sup>-</sup>] as tundra snow near Utqiaġvik (Krnavek et al 2012, *Atmos. Environ.* & P. Peterson, Pratt Lab)

## ? Blowing Snow (Antarctic model – Yang et al 2008, *GRL*)

- *Hypothesis:* wind-driven snow lofting, followed by snow sublimation & sea salt aerosol production, followed by reactions on particle surfaces to produce reactive bromine
- *Need experimental tests:* simultaneous measurements of surface snow, lofted snow grains, aerosols (size & chemistry), and reactive bromine (BrO, Br<sub>2</sub>) to connect the proposed full cycle
- In Utqiaġvik, increased wind (increased atmospheric mixing) typically is accompanied by increased O<sub>3</sub>.



# Enhanced Aerosol $\neq$ BrO Vertical Propagation

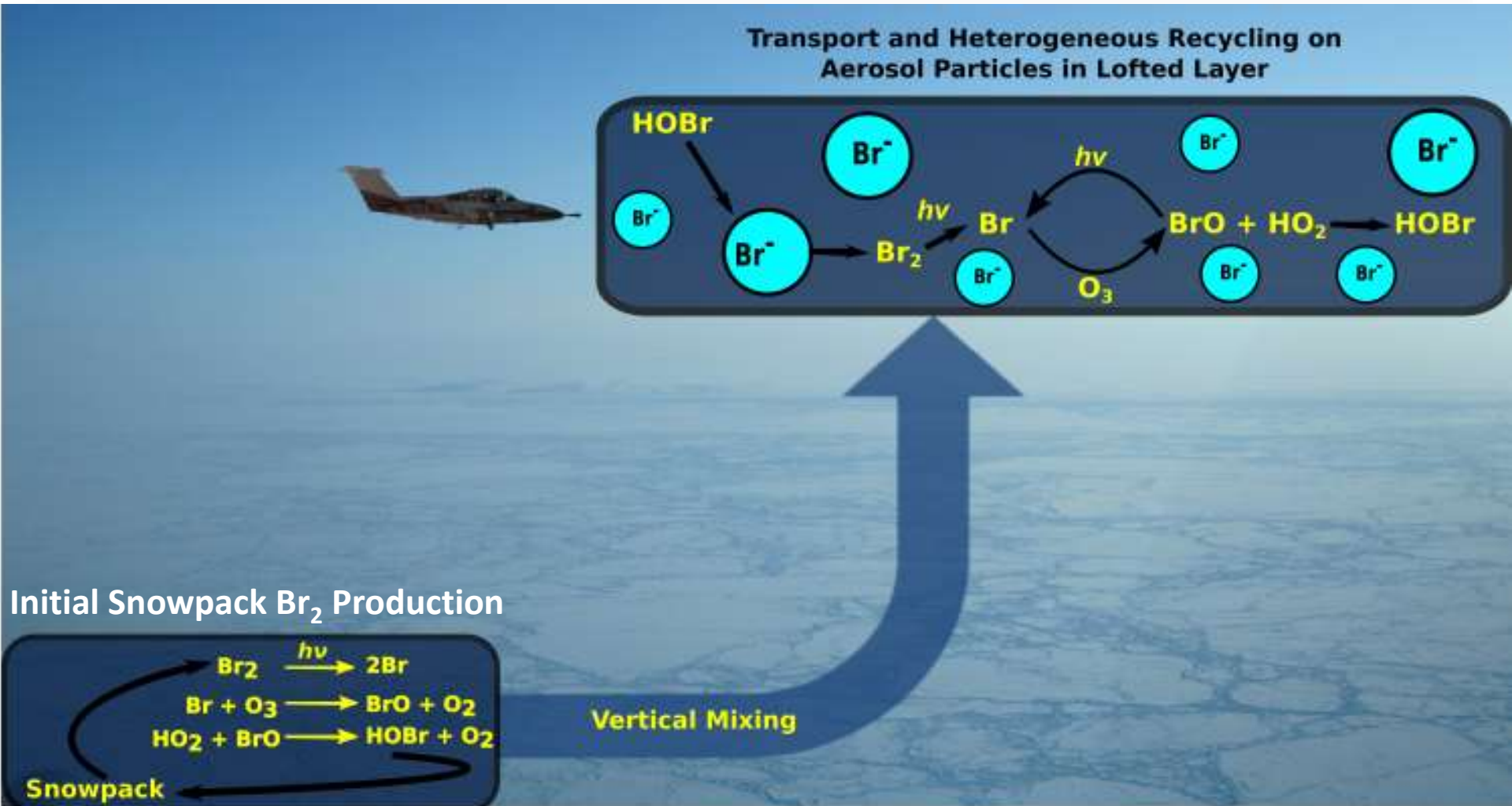


“Aerosol aloft is necessary for BrO to be present aloft, but it is not sufficient to always cause BrO to propagate vertically when aerosol extinction is present.”

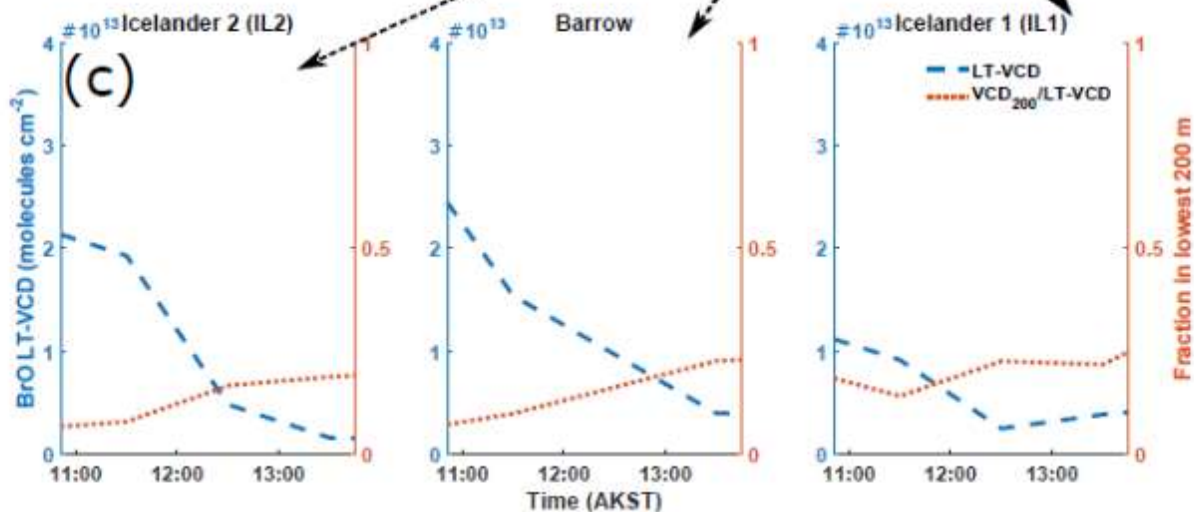
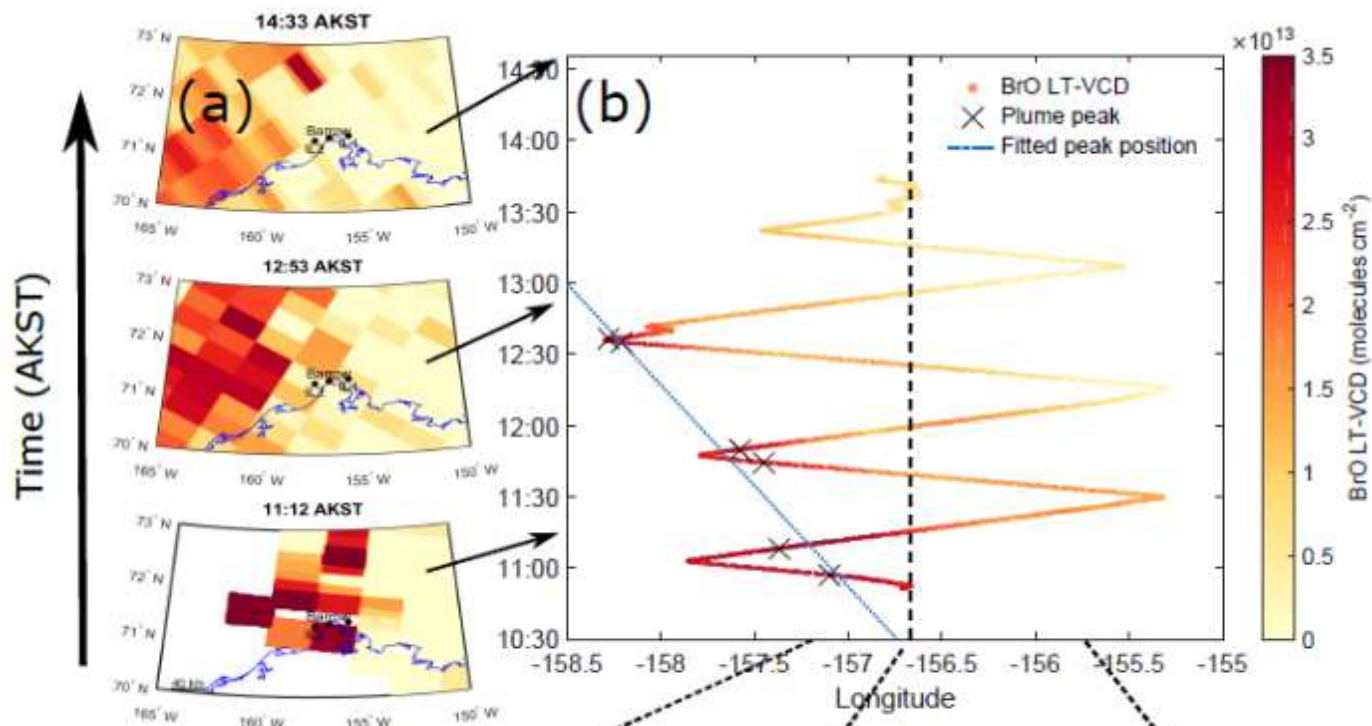
Important factors:

- Meteorology (Peterson et al 2015, *ACP*)
- Aerosol chemical composition (sea spray aerosol: aerosol  $\text{Br}^-$ )

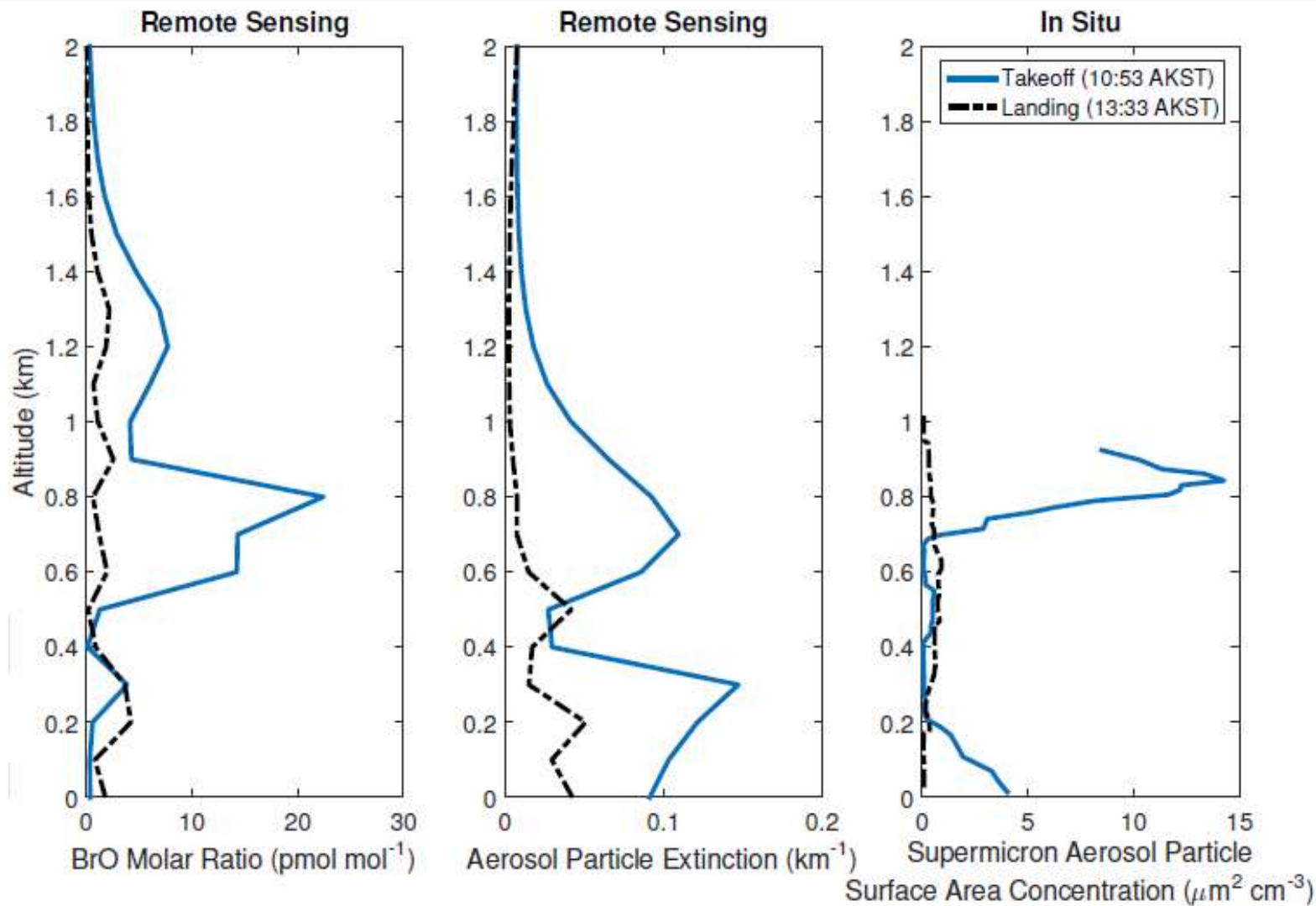
# Lofted Bromine Chemistry Sustained by Recycling Reactions on Aerosols



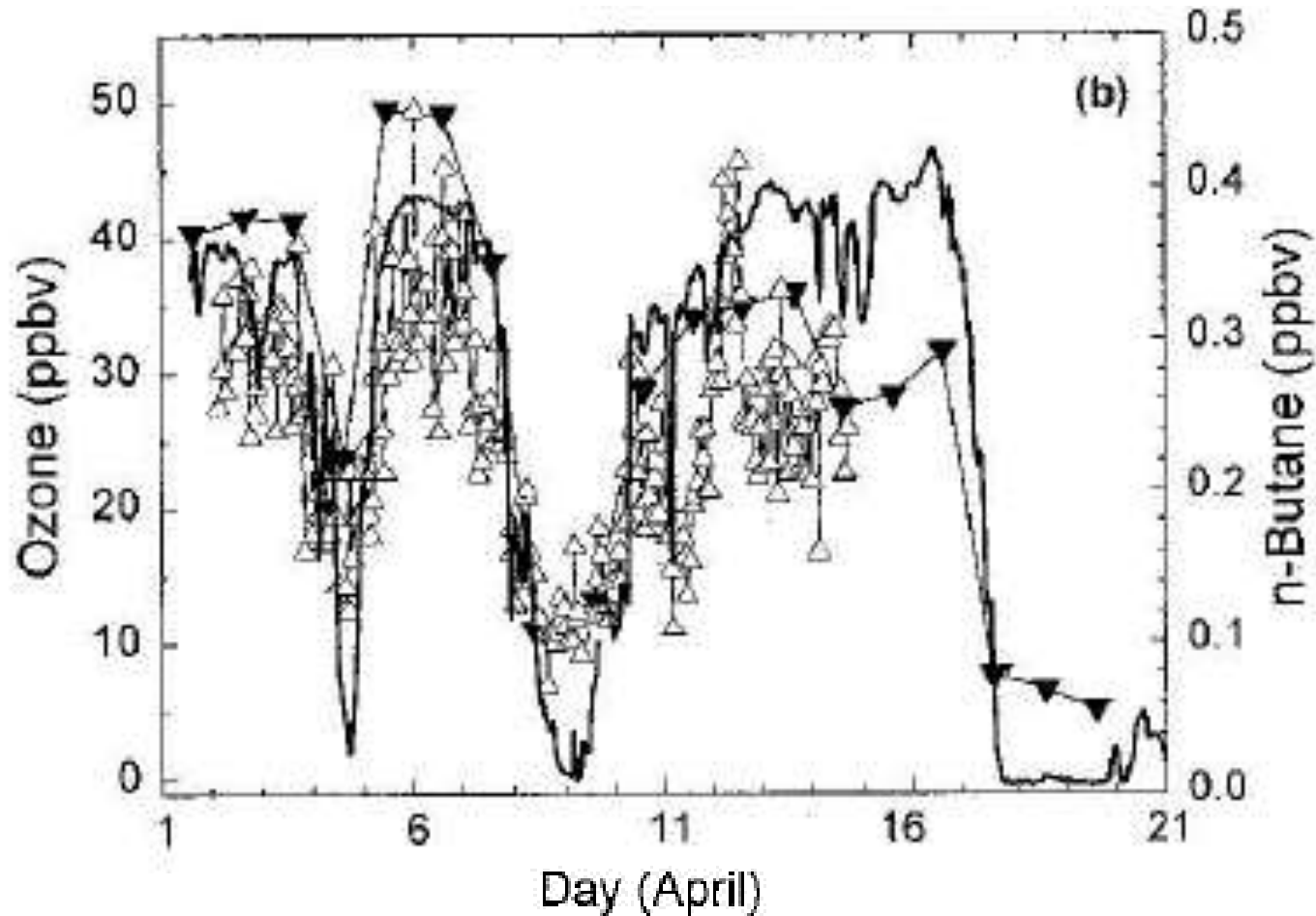
# BrO Plume moving at Speed of Air Mass



# Lofted BrO Co-located with Enhanced Supermicron Aerosol



# Evidence of Cl Atom Reactions with Hydrocarbons

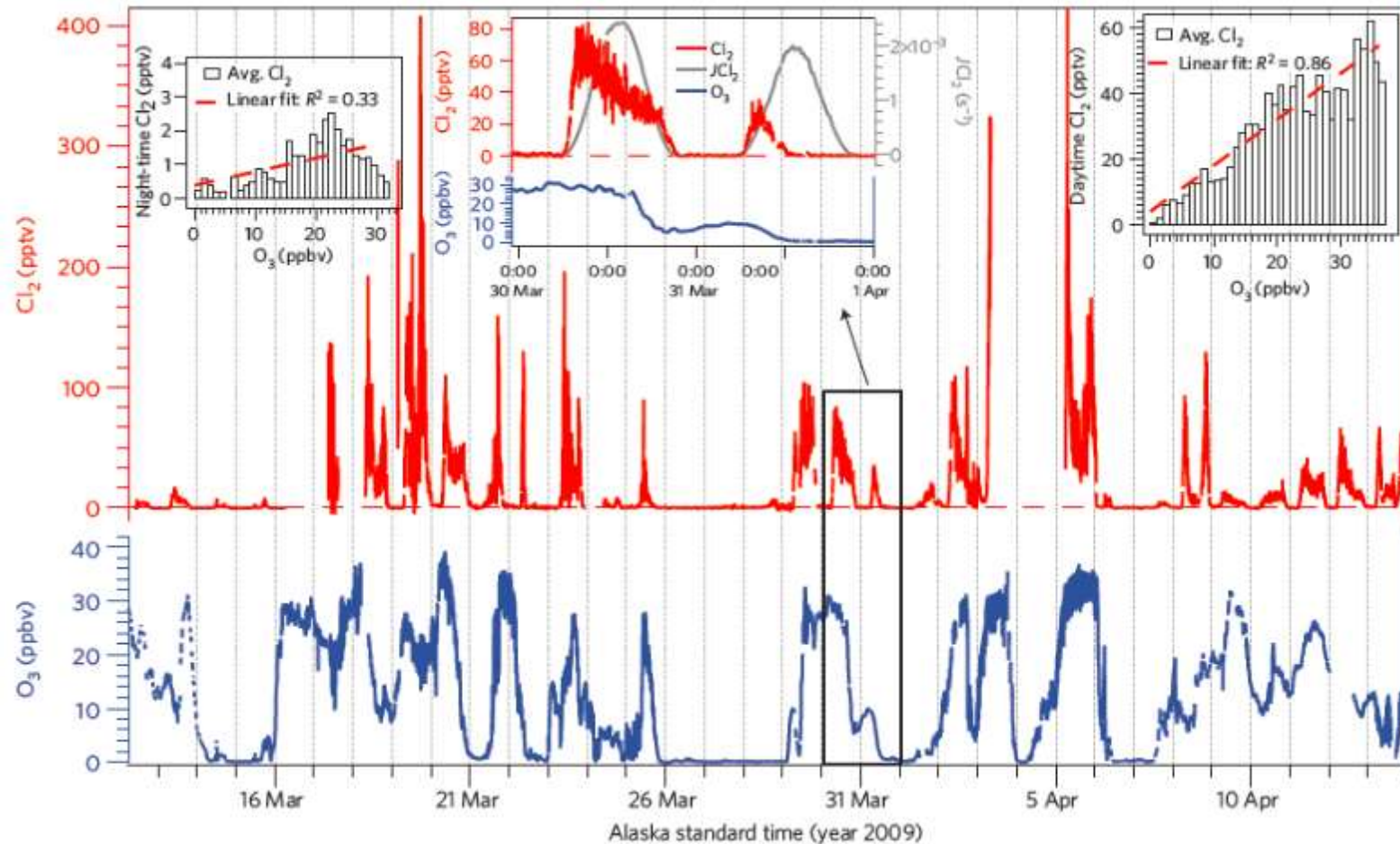


Examples:  $k_{Cl} \sim 100x k_{OH}$  for n-butane

$k_{Cl} \sim 20x k_{OH}$  for methane

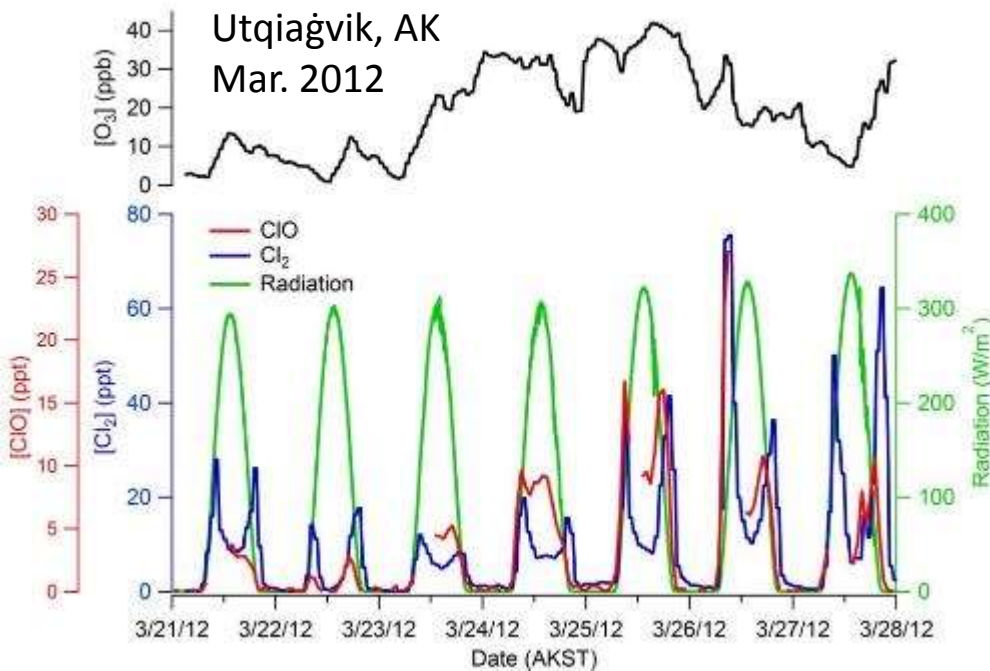
# High levels of molecular chlorine in the Arctic atmosphere

Jin Liao<sup>1,2,3</sup>, L. Gregory Huey<sup>1\*</sup>, Zhen Liu<sup>1,4</sup>, David J. Tanner<sup>1</sup>, Chris A. Cantrell<sup>5,6</sup>, John J. Orlando<sup>5</sup>, Frank M. Flocke<sup>5</sup>, Paul B. Shepson<sup>7</sup>, Andrew J. Weinheimer<sup>5</sup>, Samuel R. Hall<sup>5</sup>, Kirk Ullmann<sup>5</sup>, Harry J. Beine<sup>8</sup>, Yuhang Wang<sup>1</sup>, Ellery D. Ingall<sup>1</sup>, Chelsea R. Stephens<sup>7</sup>, Rebecca S. Hornbrook<sup>5</sup>, Eric C. Apel<sup>5</sup>, Daniel Riemer<sup>5</sup>, Alan Fried<sup>5</sup>, Roy L. Mauldin III<sup>5,6,9</sup>, James N. Smith<sup>5</sup>, Ralf M. Staebler<sup>10</sup>, J. Andrew Neuman<sup>2,3</sup> and John B. Nowak<sup>2,3</sup>



Cl<sub>2</sub> correlated with O<sub>3</sub> and radiation

# ClO Measurement – Constraint on [Cl]

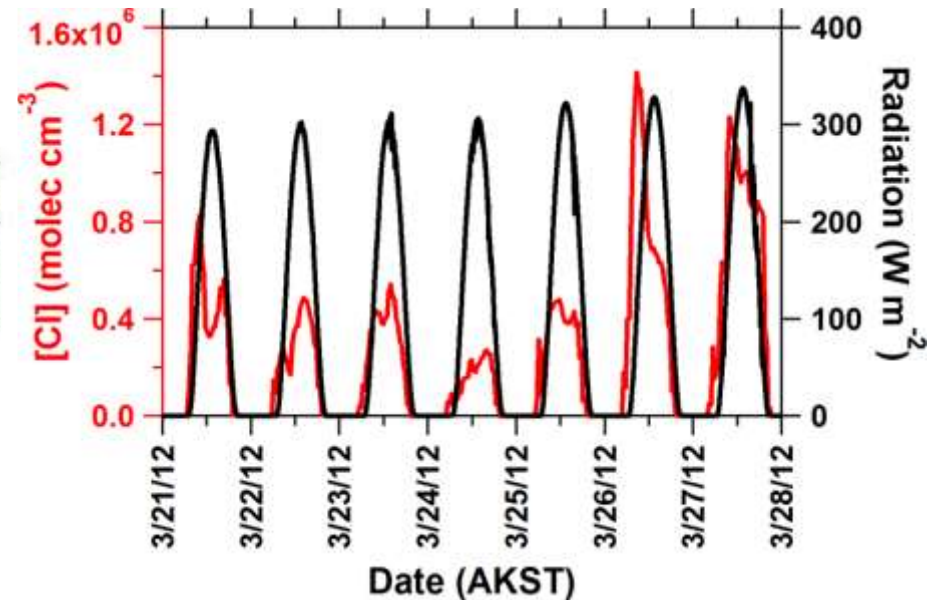


First CIMS measurement of ClO:

$3\sigma$  LOD 2.6 ppt (1 min), 0.3 ppt (1 h)

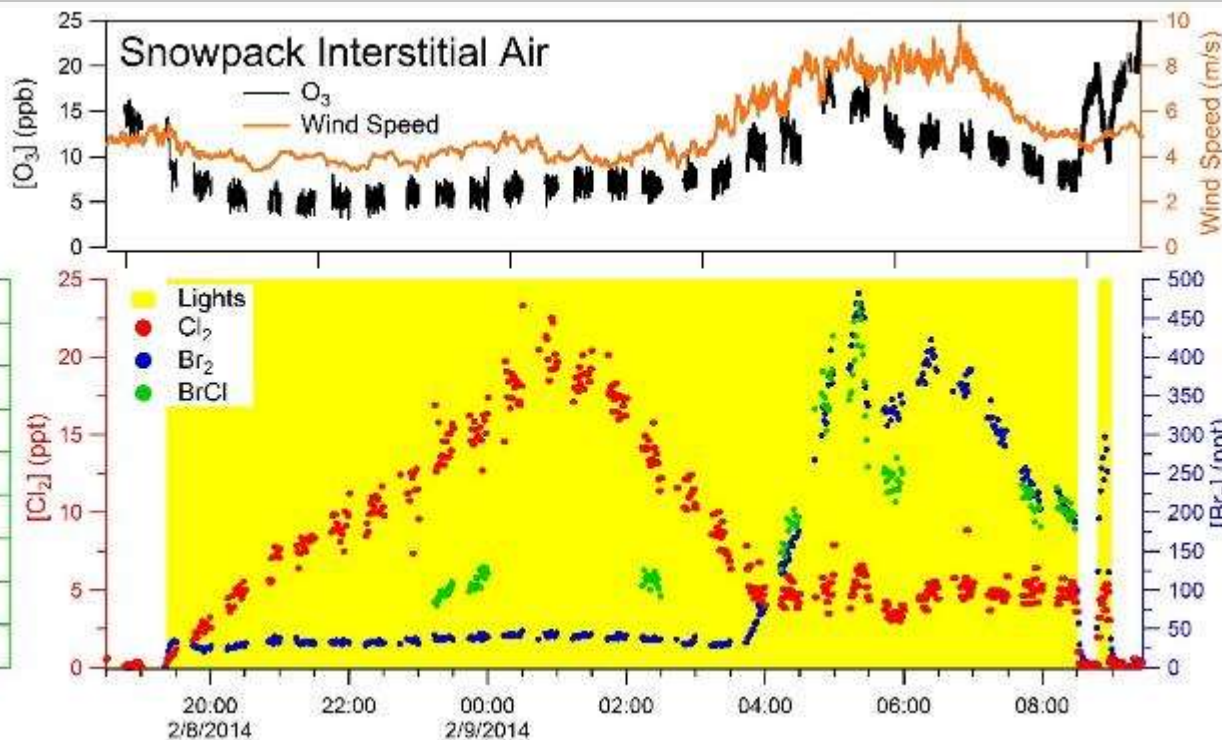
Custard et al 2016, *ES&T*

0-D model constrained by measured  $[Cl_2]$  simulates  $[ClO]$  within measurement uncertainty, providing an improved constraint on  $[Cl]$

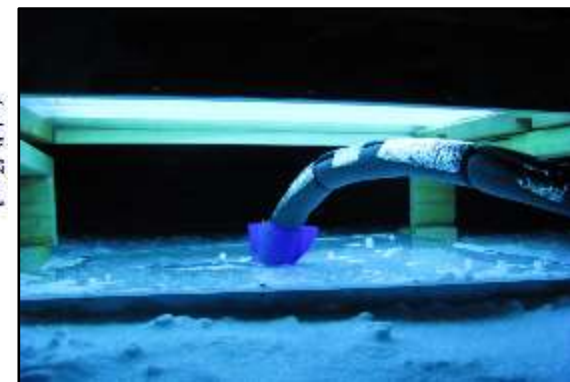


Modeled  $[Cl]$  suggest that previous calculations, based on hydrocarbon levels, are underestimates due to the longer lifetimes of the hydrocarbons

# Photochemical Snowpack $\text{Cl}_2$ and $\text{BrCl}$ Production



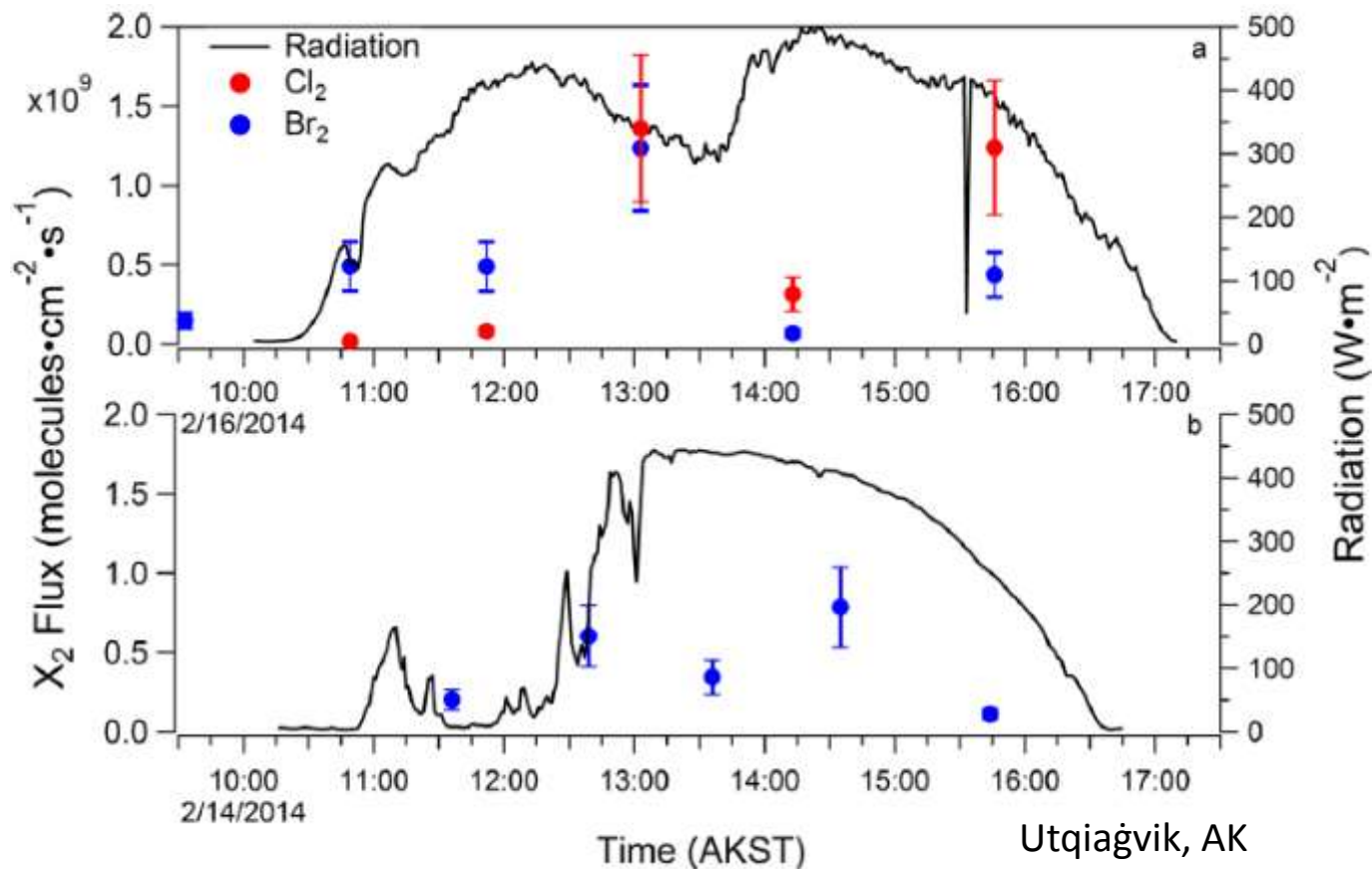
Artificially irradiated  
snowpack  
Utqiagvik, AK  
Feb. 2014



- Upon illumination, Arctic snowpack produces molecular halogens (first observations of snowpack  $\text{Cl}_2$  and  $\text{BrCl}$  production)
- $\text{Br}_2$  and  $\text{BrCl}$  followed similar trends, with production *enhanced* by the addition of  $\text{O}_3$  and at air temperatures below which  $\text{NaCl}\cdot 2\text{H}_2\text{O}$  precipitates
- $\text{Cl}_2$  production *limited* at  $<10$  ppb  $\text{O}_3$  and temperatures below which  $\text{NaCl}\cdot 2\text{H}_2\text{O}$  precipitates (consistent with previous lab studies - Sjostedt & Abbatt 2008, *ERL*)

# Snowpack Br<sub>2</sub> and Cl<sub>2</sub> Fluxes

First Br<sub>2</sub> and Cl<sub>2</sub> snowpack fluxes calculated based on vertical concentration gradients above sunlit snowpack

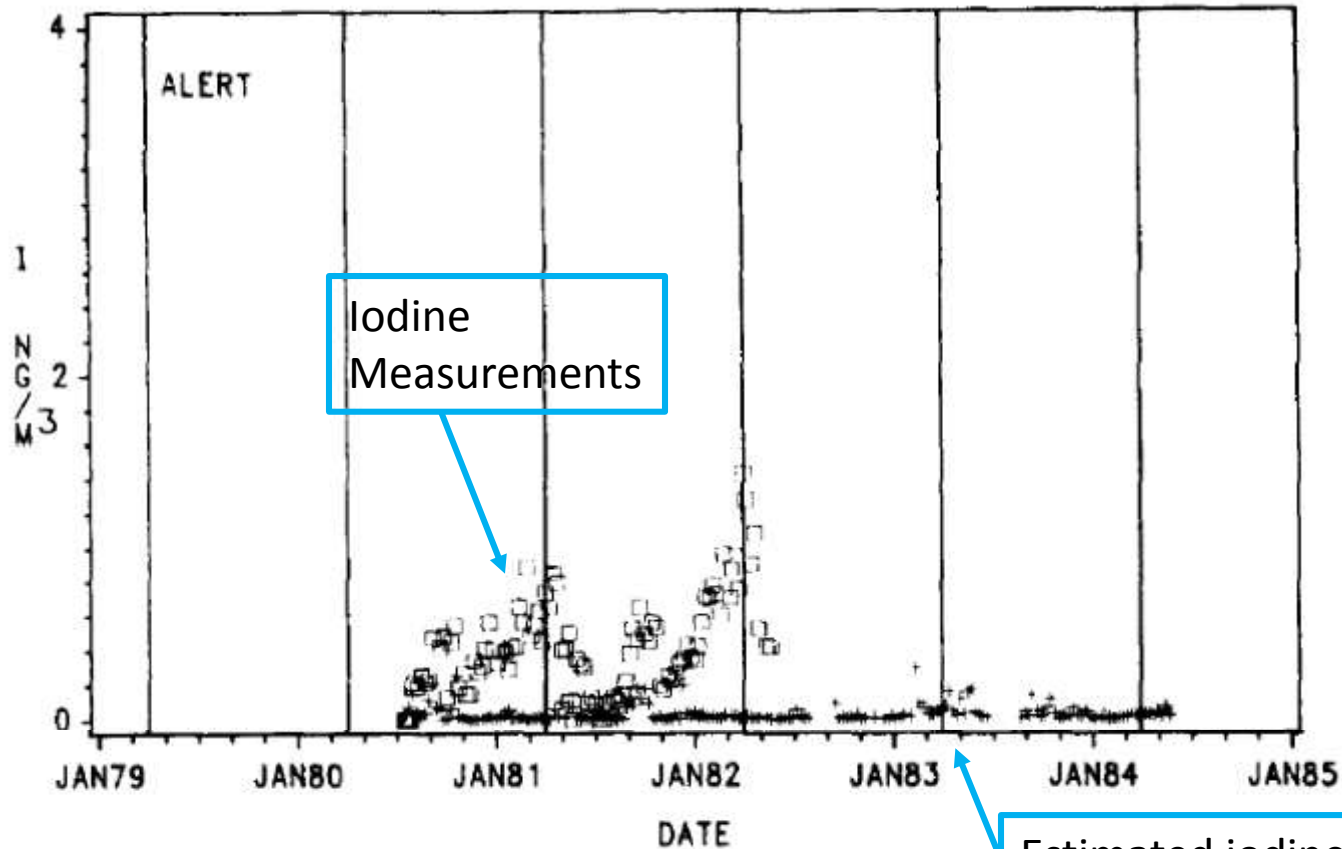


Consistent with previous modeling studies estimating Br<sub>2</sub> fluxes:

$9.0 \times 10^7$  to  $2.7 \times 10^9$  molecules cm<sup>-2</sup> s<sup>-1</sup>

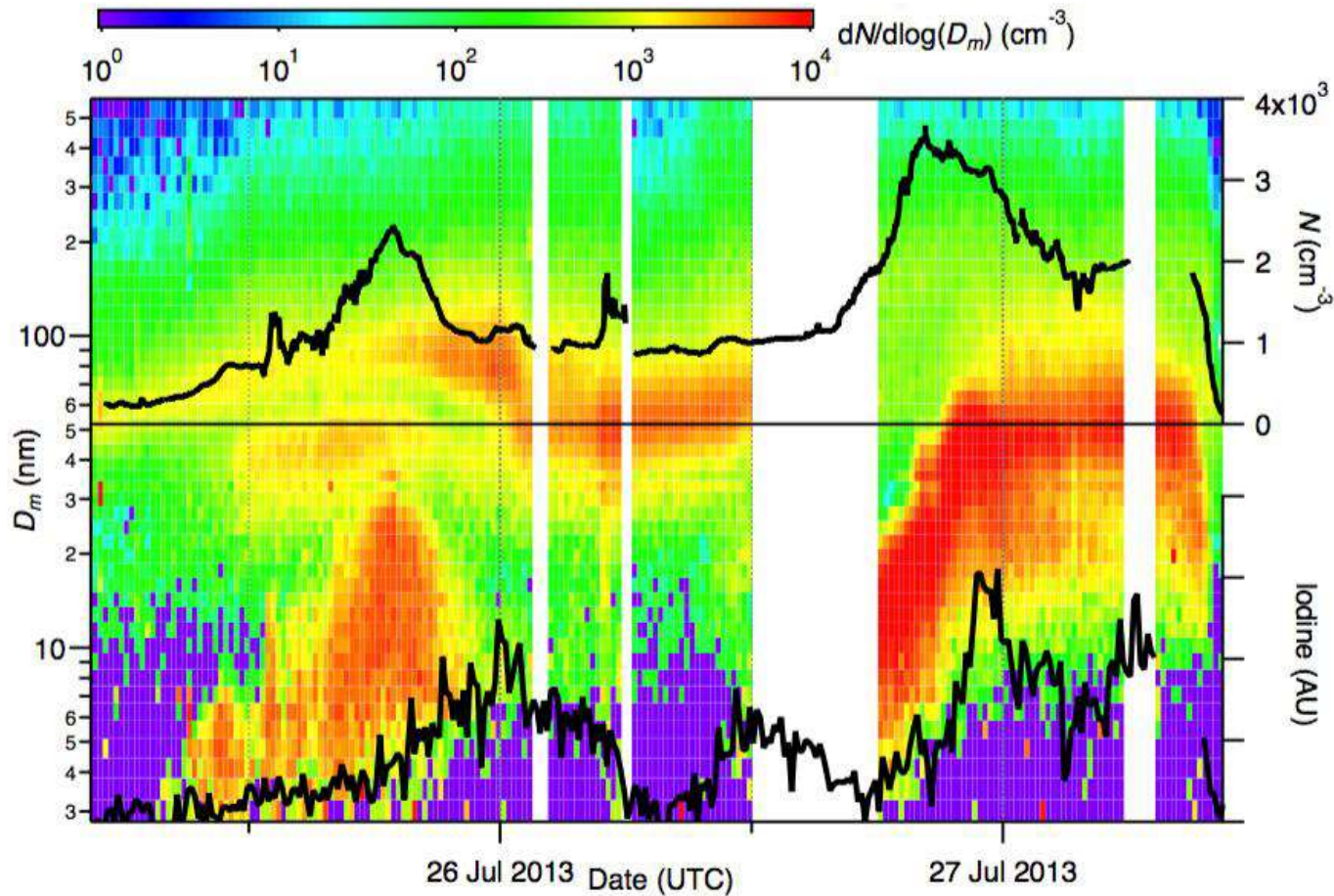
# Evidence of Arctic Iodine Chemistry

## Particulate Iodine

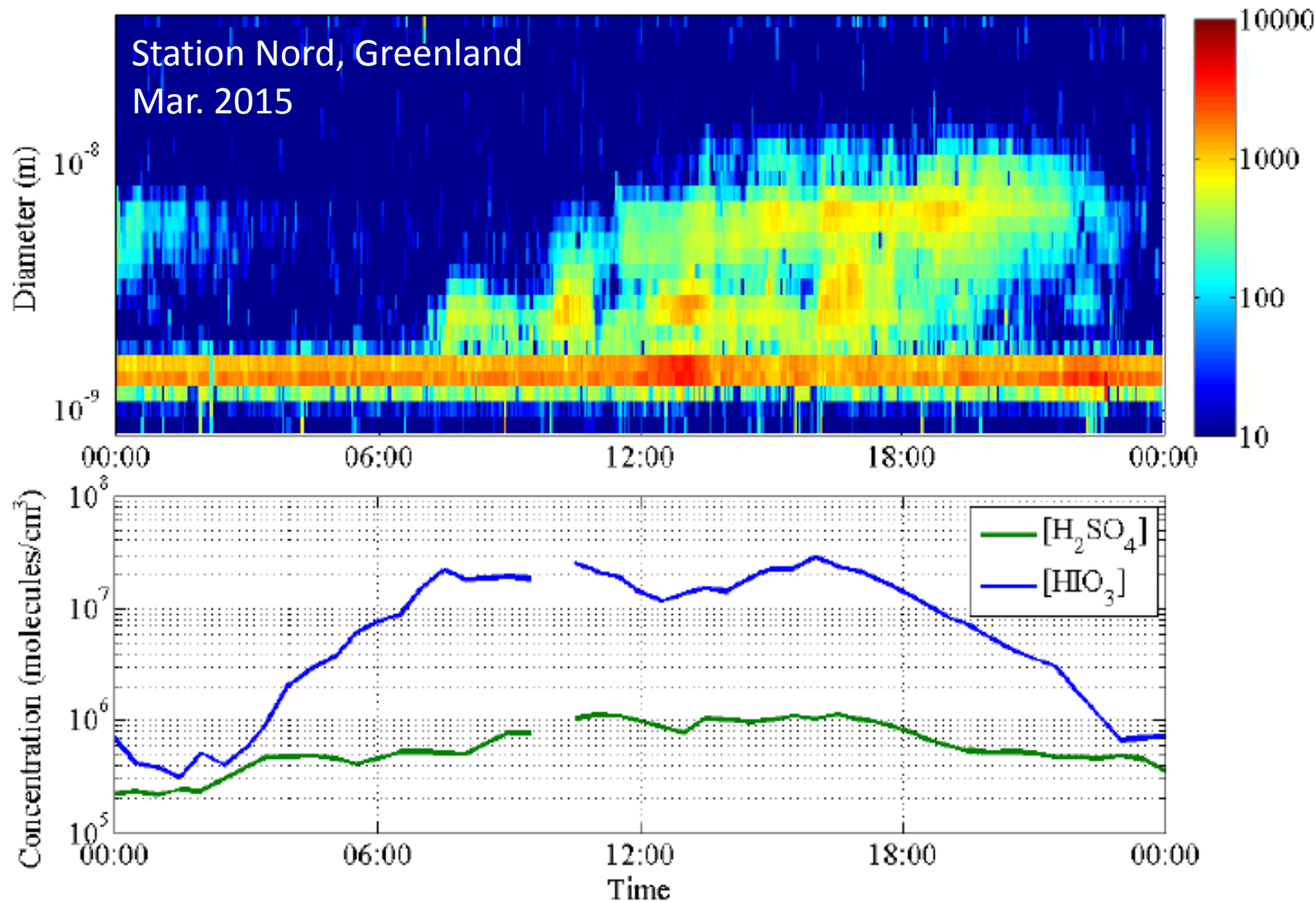


Estimated iodine based on sea salt & crustal material

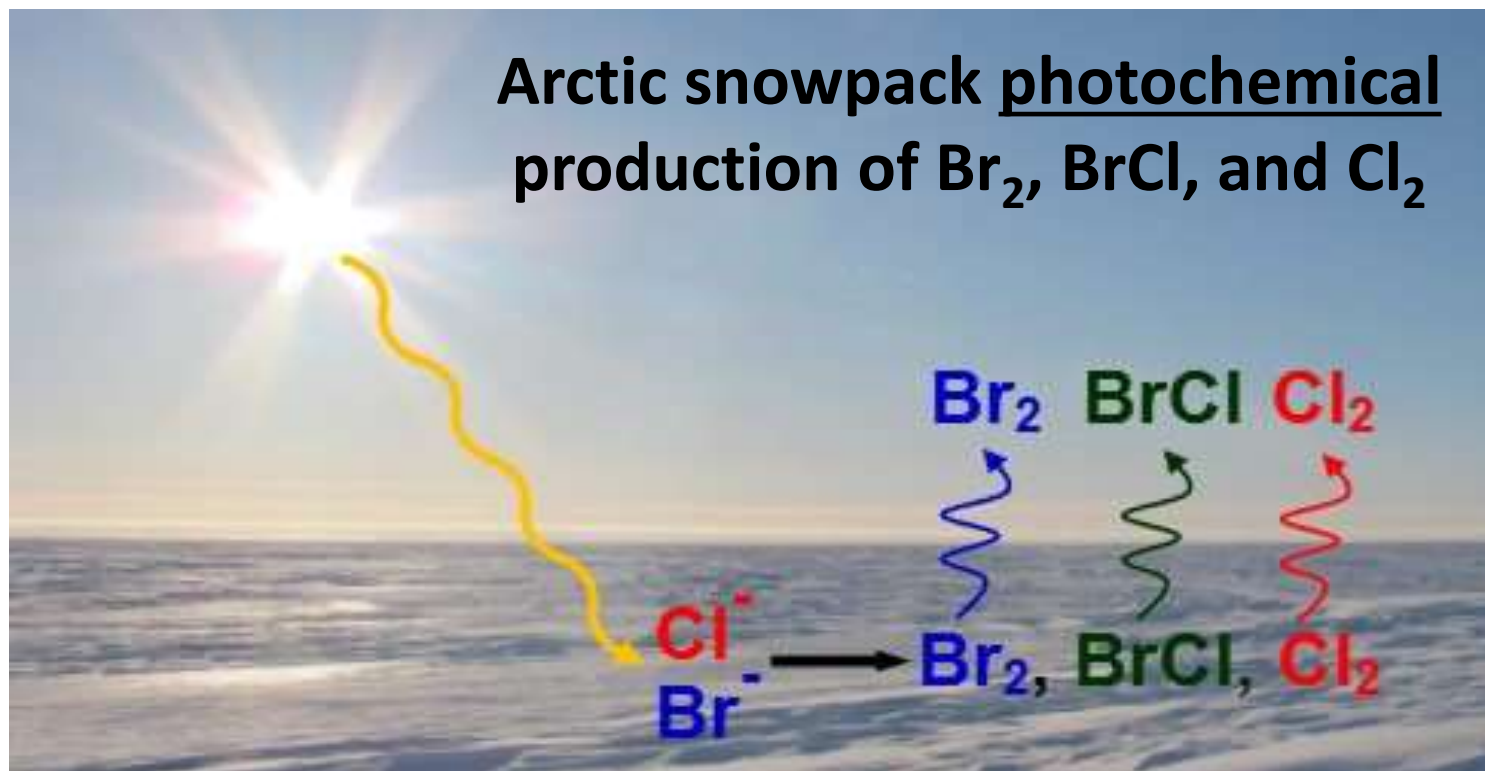
# Aerosol Iodine Associated with New Particle Formation and Growth



# Iodic acid ( $\text{HIO}_3$ ) Clusters Associated with New Particle Formation



# Snowpack Molecular Halogen Production



- Relative amounts of  $\text{Br}_2$ ,  $\text{BrCl}$ , and  $\text{Cl}_2$  may be related to snowpack temperature & precipitation of  $\text{NaCl} \cdot 2\text{H}_2\text{O}$
- Constraint on  $[\text{Cl}]$  provided by simultaneous  $\text{Cl}_2$  and  $\text{ClO}$  CIMS measurements

Custard et al. 2016, *ES&T*

Custard et al. 2017, *ACS Earth & Space Chem.*